Secondary structure, orientation and oligomerisation of phospholemman, a cardiac transmembrane protein

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Abstract – Human Phospholemman (PLM) is a 72 residue protein, which is expressed at high density in the cardiac plasmamembrane and in various other tissues. It forms ion channels selective for K^+, Cl^- and taurine in lipid bilayers and co-localises with the Na^+/K^+ ATPase and the Na^+/Ca^{2+} exchanger, which may suggest a role in the regulation of cell volume. Here we present the first structural data based on synthetic peptides representing the transmembrane domain of PLM. Perfluoro-octaneoate-PAGE of reconstituted proteoliposomes containing PLM reveals a tetrameric homooligomerisation. Infrared spectroscopy of proteoliposomes shows that the PLM peptide is completely α-helical even beyond the hydrophobic core residues. Hydrogen/deuterium exchange experiments reveal that a core of 20 – 22 residues is not accessible to water, thus embedded in the lipid membrane. The maximum helix tilt is (17±2)° obtained by attenuated total reflection infrared spectroscopy. Thus our data supports the idea of ion channel formation by the PLM transmembrane domain.

Key words: secondary structure, oligomerization, membrane proteins, infrared spectroscopy, peptide, phospholemman.

Phospholemman (PLM) is a 72-residue protein which is found in the central nervous system, skeletal muscle and at high density in the cardiac sarcolemma – the plasma membrane which surrounds the heart muscle fibre – from which it was first purified from (Palmer et al. 1991). It is phosphorylated by protein kinase A and C at different sites in the cytoplasmic domain (Palmer et al. 1991), which suggests that PLM may be an important control point in the function of heart cells. The biological importance of PLM is further emphasised by the fact that PLM exhibits a close similarity to a number of other newly identified membrane proteins, including CHIF, RIC, and MAT-8, that form a new superfamily of membrane proteins with a single transmembrane domain (Sweadner and Rael 2000).

The function of PLM is not completely understood, yet recent evidence suggests that PLM may function as a regulator of Na^+/ K^+-ATPase (Silverman et al. 2005) and that it also co-localises with the Na^+/Ca^{2+}-exchanger (Mirza et al. 2004). Electrical measurements of PLM in artificial lipid bilayers and oocytes show that PLM facilitates the membrane flux of ions (Chen et al. 1999) and taurine transport (Moorman et al. 1995). Based on these findings it has been suggested that PLM has a function in the regulation of cell volume either as a modulator of a swelling-activated signal transduction pathway or directly facilitating osmolyte influx (Davis et al. 2004).

To date, there is no structural data available about PLM based on physical measurements, however a solid-state NMR study of the related MAT-8 and CHIF proteins show that they have an α-helical transmembrane domain (Franzin et al. 2004). On the basis of hydropathy analysis of the amino acid sequence, PLM is predicted to have a single transmembrane domain from residue 18 to 37 (Palmer et al. 1991). The C-terminus is located intracellularly, which has been shown by resistance to protease digestion, while the N-terminus is located extracellulary as shown by antibody labelling (Chen et al. 1998). Here we show experimentally that PLM has indeed a transmembrane domain, which forms a tetramer of α-helices in a lipid bilayer membrane. The α-helices have a predominant transmembrane orientation.

Results

Gel-electrophoresis

SDS-PAGE of the lipid vesicle reconstituted PLM peptide shows a single band depicted in figure 1B. A plot of the relative migration distance of the marker proteins against the logarithm of the molecular mass is used to determine the molecular mass of PLM as 3.9±0.5 kDa. The expected molecular mass of the PLM peptide monomer is 3.04 kDa. Replacing the detergent SDS with perfluoro-octaneoate (PFO) shows again a single band at the molecular mass of 12.6±0.5 kDa (figure 1A).

Infrared spectroscopy

The amide I infrared absorption band of the lipid vesicle reconstituted PLM peptide centered at 1658 cm^{-1} is characteristic of predominantly α-helical structure in terms of wavenumber and bandshape (figure 2A). To characterise the secondary structure quantitatively a polarisation independent spectrum has been generated by combining the spectra at parallel and perpendicular polarisation (eq. 1). After applying the band narrowing technique of Fourier self-deconvolution the spectrum can be fitted by six Gaussian curves as shown in figure 2B. The secondary structure composition is obtained from the relative area under each curve as 82% α-helix, 11% ß-sheet and 7.3 % disordered structure. A detailed assignment of each band is presented in table 1 (Tamm and Tatulian 1997). Furthermore a second peptide with the residue Leu16 labeled with $^{13}C^{18}O$ at its carbonyl group shows a small absorption band at 1595 cm^{-1} (figure 3), which is consistent with this residue beeing in an α-helical environment (Torres et al. 2001).

For orientational analysis the dichroic ratio of the amide I band is calculated from the integrated amide I absorption at parallel and perpendicular polarised light (eq. 2) as R = 3.86±0.1. Using equation 3 and 4 the tilt angle of the helix with respect to the bilayer normal is calculated as β = 17°±2°.

The polarisation independent absorption spectrum after 9 h of incubation with D$_2$O saturated nitrogen shown in figure 3 (dotted line) reveals only a small amount of exchange. Integration of the amide II band yields an exchange of 24±3 %, which remains constant between 6 to 24 hours. Please note that a peptide with the residue Leu16 labeled with $^{13}C^{18}O$ at its carbonyl group has been used for the deuterium-exchange study, which does not affect the results.

Discussion

The method of PFO-PAGE has been developed recently to study the oligomerisation of transmembrane protein complexes (Ramjeesingh et al. 1999) and has been used successfully for synthetic peptides (Therien and Deber 2002). For a PLM transmembrane peptide we find M_r = 12.6 kDa, which corresponds to a tetrameric oligomerisation (n=4.1), given that the monomer mass is M_r = 3.04 kDa. Higher and lower oligomerisation states would be clearly resolved on the gel (figure 1A) but no other distinctive bands are visible. Thus we can safely conclude that the PLM peptide forms a tetramer in lipid vesicles. The oligomerisation is clearly disrupted by SDS as shown in figure 1B by the appearance of one band at M_r = 3.9 kDa. The difference to the expected peptide molecular mass of 3.04 kDa reflects the inaccuracy of gel electrophoresis to detect this low molecular mass compound; the accuracy based on the standard error is however sufficient to distinguish between monomer, dimer, trimer and higher oligomerisation species.

The secondary structure analysis reveals that the peptide is mainly α-helical in a lipid bilayer. The small β-sheet content of 11% reflects most likely unspecific aggregation of peptide, which did not insert into the lipid bilayer, a phenomenon which is commonly observed with hydrophobic peptide preparations (Kukol and Arkin 2000). Taking into account only the lipid bilayer inserted amounts of 82% α-helix and 7.3% disordered structure , we can derive that 26 aminoacid residues are in an α-helical conformation, while 2 residues are in a disordered random coil conformation.

The helix tilt angle of β = 17°±2° reveals that the peptide has a predominantly transbilayer orientation as opposed to an orientation parallel to the membrane plane. It should be stressed however, that the angle of 17° reflects a maximum tilt angle assuming that the sample is perfectly ordered as pointed out by Arkin et al. (Arkin et al. 1997). The actual tilt angle is likely to be lower, because perfect sample order cannot be achieved with the present method of preparation. The fraction of ordered sample is typically between 0.6 and 0.8 as obtained by site-specific dichroism analysis of multiply labelled samples for a similar system (Kukol et al. 1999). The tilt angle taken together with the calculated length of the helical segment strongly indicates that the helix segment extends beyond the lipid bilayer into the solution phase. Embedding an α-helix of 26 residues in a lipid bilayer would require higher tilt angles. Alternatively the local thickness of the bilayer could be slightly larger to account for

any hydrophobic mismatch, but given that there is a strong hydrophobic core of only 19 aminoacid residues this possibility seems unlikely. From the amount of hydrogen/deuterium exchange of 24±3% (fig.3) we obtain that 7±1 residues are accessible to water. That means that the lipid membrane embedded part is 20-22 residues long.

Based on the considerations above, we present the following tentative model of the PLM transmembrane domain (figure 4), a tetramer of α-helices tilted by 17° against the bilayer normal. The helix extends beyond the hydrophobic transmembrane core of 20 or more aminoacid residues (shown in light grey, figure 4) to at least 26 residues. This model explains the experimental findings of ion channel activity of PLM (Kowdley et al. 1997; Chen et al. 1999; Kadowaki et al. 2004) in the way that a tetramer is able to form a hydrophilic pore lined by the carbonyl groups of the peptide backbone through the lipid bilayer. While the role of PLM has been mainly discussed in terms of an efflux pathway the short extracellular domain shows a remarkable resistance against proteases (Chen et al. 1998), which is supported by our finding that the α-helical structure extends into the extracellular space, possibly forming a well defined structure for the entrance of ions or osmolytes. While our model sheds light on the proposed function of PLM as a transmembrane channel permeable to ions and small molecules (Davis et al. 2004) the function as a regulator of the Na^+/K^+-ATPase or the Na^+/Ca^{2+}-exchanger found in cardiac and skeletal muscle cells remains unexplained. It should however be noted that long-term cell volume regulation is due to the maintanance of ion gradients by the Na^+/K^+-ATPase (Lang et al. 1998).

Conclusion

This work presents the first experimental evidence that PLM is α-helical and inserts into lipid membranes in a trans-bilayer fashion forming a tetramer. Further experiments are necessary to resolve the detailed transmembrane structure, as it is possible with the method of site specific infrared dichroism (Kukol 2005). The ion channel properties of PLM have been studied so far with full length PLM and truncated version (residues 1-43). Given that the C-terminus 1-18 is highly protease resistant, it may have a substantial effect on the channel properties. Electrical conductance measurements of the transmembrane domain on its own would help to understand the function of the C-terminal domain. While an increasing amount of

structural and functional information is emerging significantly more experimental data is necessary to understand this seemingly simple protein.

Materials and Methods

Peptide purification and reconstitution: Two synthetic peptides corresponding to the predicted transmembrane domain of human phospholemman of the sequence DYQSLQIGGLVIAGIL*FI-LGILIVLSRR (M_r = 3042) was obtained by standard solid-phase FMOC-chemistry (ABC, Imperial College, London). The second peptide contained a $^{13}C^{18}O$ carbonly group labeled Leu residue at the position 16 indicated with a star, but it is otherwise chemically identical. The Leu label was produced from 1-^{13}C labeled Leu (Cambridge Isotopes, Cambridge MA) by acid catalysed oxygen exchange in 95% enriched $H_2^{18}O$ (Campro Scientific, Nethelands) as decribed previously (Torres et al. 2001). The peptides were further purified as described elsewhere (Kukol et al. 1999) for analogous transmembrane peptides. Briefly, the peptide was dissolved in trifluoroacetic acid and purified by reversed phase chromatography (Jupiter 5C4-300 Å column, Phenomenex, Cheshire, UK). Peptide elution was achieved with a linear gradient to a final solvent composition of 5 % H_2O, 38% acetonitrile and 57% 2-propanol. All solvents contained 0.1% (v/v) trifluoroacetic acid. The correct molecular mass of the peptide in each fraction was checked by positive ion electrospray mass spectrometry. After lyophilisation of the pooled fractions the peptide was dissolved in 1,1,1,3,3,3-hexafluoro-2-propanol, containing the lipid dimyristoylphosphocholine (Sigma, UK), to a peptide to lipid ratio of 1:25 (w/w). The organic solvent was removed by rotational evaporation under vacuum at room temperature. Lipid vesicles were formed by adding water to a lipid concentration of 10 mg/ml and gentle rotation of the flask for 20 min followed by 2 min sonication. Subsequently the mixture was frozen and thawed for three times to achieve a homogenous peptide/lipid distribution and stored frozen for further analysis.

Gel electrophoresis: The standard tris-tricine peptide separation method of Schagger and von Jagow was employed (Schagger and von Jagow 1987). Samples were run on pre-cast tris-tricine 10-20% polyacrylamide gradient gels obtained from Bio-Rad Laboratories. Depending on the gel in question, sample buffers contained either 4% w/v SDS or 4% w/v perfluoro-octaneoate (PFO). Likewise, cathode running buffers contained either 0.1% w/v SDS or PFO. Lipid-peptide samples were mixed 1:1 with sample buffers, centrifuged at 10,000 rpm for 5 minutes and then loaded on the gel (Ramjeesingh et al. 1999). For SDS-PAGE, the sample was heated to 95ºC for

5 minutes prior to mixing with sample buffers. Gels were run at a constant voltage of 50V. In the case of PFO-PAGE, the gel was run in a controlled temperature environment of approx. 4ºC, in order to maintain weakly associated complexes. Molecular mass markers were obtained from Sigma Laboratories (UK). Gels were stained with colloidal coomassie blue. The relative mobility R_f of a band was obtained by dividing the distance for the particular band by the maximum distance for the sample buffer.

Infrared Spectroscopy: Fourier transform infrared spectra (FTIR) were recorded on a Bruker Tensor spectrometer (Bruker Optics, Coventry UK) equipped with a high sensitivity liquid nitrogen cooled MCT/A detector and purged with dry air at a dew point of -73ºC (Parker Purge Gas Generator, Haverhill, MA). Attenuated total reflection (ATR) – FTIR spectra were measured with a 25 reflections ATR accessory from Graseby Specac (Kent, UK) and a wire grid polariser (0.25 µm, Graseby Specac). 100 µL of the peptide/lipid solution in hexafluoroisopropanol was dried onto a germanium trapezoidal internal reflection element (50 x 2 x 20 mm) under a stream of nitrogen. After the solvent was removed, the lipid/protein film was hydrated for 12h with a water saturated stream of nitrogen. Dichroic ratios similar to lipid vesicle solutions in water were achieved through this procedure (data not shown). After removal of excessive water the FTIR spectrum of a semi-dried lipid film was obtained from 1000 averaged interferograms and 1 point zero filling Happ-Genzel apodisation. Absorption spectra were recorded at parallel ($A_{parallel}$) and perpendicular polarisation ($A_{perpendicular}$). Typically three independent measurements were performed.

For hydrogen/deuterium exchange the lipid/protein film was subjected to a stream of deuterium oxide saturated nitrogen for up to 24h, while spectra were measured every three hours.

Data analysis: For secondary structure analysis a polarisation independent spectrum was obtained by combining the spectra at parallel and perpendicular polarisation (Marsh 1999):

$$A = A_{parallel} + 1.44 A_{perpendicular} \qquad (1)$$

A spectrum obtained at one polarisation alone cannot be used, because intensities may depend on orientation rather than the amount of a specific secondary structure element. The so-obtained polarisation independent spectrum is equivalent to a transmission spectrum obtained without polarisation.

Fourier self-deconvolution (Kauppinen et al. 1981) was applied to the amide I region of the spectrum in order to resolve the fine structure of the absorption band. The enhancement factor used was 2.0 and the half-height bandwidth was set at 13 cm^{-1}, as reported previously for secondary structure analysis (Byler and Susi 1986).

For orientational analysis the dichroic ratio, R, was calculated as the ratio between the integrated absorption A of parallel and perpendicular polarised light of the amide I absorption band between 1670 and 1645 cm^{-1}:

$$R = \frac{A_{parallel}}{A_{perpendicular}} \qquad (2)$$

The dichroic ratio is related to the order parameter S (Fringeli and Gunthard 1981):

$$S = \frac{E_x^2 - RE_y^2 + E_z^2}{E_x^2 - RE_y^2 - 2E_z^2} \left(\frac{3\cos^2 \alpha - 1}{2} \right)^{-1} \qquad (3)$$

whereby $\alpha = 39°$ is the angle between the molecular director and the transition dipole moment of the amide I vibration mode (Marsh et al. 2000) and E_x, E_y, E_z are the electric field components of the evanescent field given by Harrick (Harrick 1967) as $E_x = 1.399$, $E_y = 1.514$, $E_z = 1.621$ assuming that the thickness of the deposited film is much larger than the penetration depth of the evanescent field. For α-helical peptides the order parameter S can be used to calculate the tilt angle β between the helix and the axis normal to the membrane plane (Fringeli and Gunthard 1981):

$$S = \frac{3 < \cos^2 \beta > -1}{2} \qquad (4)$$

The helix tilt obtained in this way is only a maximum value, assuming that the protein sample is completely ordered, as pointed out by Arkin (Arkin et al. 1997).

In order to analyse the hydrogen/deuterium exchange the amide II band between 1525 cm^{-1} and 1570 cm^{-1} of the polarisation independent spectrum was integrated, and its area A_{II} was corrected by dividing by the total amide I area ($y = A_{II}/A_I$) in order to account for changes in band intensities due to swelling of the sample. The amount of D_2O exchange was then calculated by dividing the corrected amide II area in D_2O by the corrected amide II area in H_2O.

Acknowledgements

Competing Interests

None.

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Tables and Figures

Tables

Table 1: Secondary structure analysis of the lipid vesicle reconstituted PLM peptide. The assignment follows the review by Tatulian (Tatulian 2003) and references therein.

Absorption peak centre/cm^{-1}	Fraction of total area	Assignment
1684	6.4%	β-sheet
1658	82%	α-helix
1644	3.1%	Disordered
1635	4.6%	β-sheet
1625	1.3%	Disordered
1616	2.9%	Disordered

Figure legends

Figure 1: A) Top: Graph of the relative mobility R_f of the marker proteins in perfluorooctaneoate (PFO) -PAGE in dependence of the logarithm of molecular mass. The arrow indicates the relative mobility of the phospholemman (PLM) peptide. Bottom: PFO-PAGE gel showing marker proteins (left lane) and PLM (right lane). B) Same as A) for SDS-PAGE. The molecular mass of the marker proteins is 16.9 kDa, 14.4 kDa, 10.6 kDa, 8.2 kDa, 6.2 kDa, 3.5 kDa and 2.5 kDa. The vertical bars indicate the standard error of the measurement.

Figure 2: A) Attenuated total reflection (ATR) -FTIR absorption spectra of the amide I/amide II region of lipid vesicle reconstituted PLM. The solid curve shows the spectrum obtained at a polarisation parallel to the plane of incidence; the dashed curve shows the spectrum obtained at a

polarisation perpendicular to the plane of incidence. B) The polarisation independent FTIR absorption spectrum after Fourier self-deconvolution (solid line). The dashed curve shows the individual component Gaussian curves used for curve-fitting the spectrum.

Figure 3: Polarisation independent FTIR absorption spectra of lipid reconstituted PLM from H_2O (solid line) and after 9 h of incubation with D_2O saturated nitrogen gas (dashed line).

Figure 4: Model of the PLM transmembrane domain, representing α-helices as cylinders. The light grey shade highlights the hydrophobic aminoacid residues, while the hydrophilic residues are shaded in dark grey. The approximate position of the lipid bilayer is indicated. The top part of the figure represents the extracellular and the bottom part the intracellular region. The N-terminal Asp (top) and C-terminal Arg residues of the transmembrane peptide oligomer are highlighted. This figure was created with VMD (Humphrey et al. 1996).