

1 **Simplified Immobilisation Method for Histidine-tagged Enzymes in**
2 **Poly(methyl methacrylate) Microfluidic Devices**

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1 Abstract

2 Poly(methyl methacrylate) (PMMA) microfluidic devices have become promising
3 platforms for a wide range of applications. Here we report a simple method for
4 immobilising histidine-tagged enzymes suitable for PMMA microfluidic devices.
5 The reported 1-step-immobilisation is based on the affinity of His-tag/Ni-NTA
6 interaction and does not require prior amination of PMMA surface, unlike many
7 existing protocols. We compared it with a 3-step immobilisation protocol involving
8 amination of PMMA and linking NTA *via* a glutaraldehyde cross-linker. These
9 methods were applied to immobilise transketolase (TK) in PMMA microfluidic
10 devices. Binding efficiency studies showed that approximately 15% of the supplied
11 TK was bound using the 1-step method and approximately 26% of the enzyme bound
12 by the 3-step method. However, the TK-catalysed reaction producing L-erythrulose
13 performed in microfluidic devices showed that specific activity of TK in the device
14 utilising 1-step immobilisation method was approximately 30% higher than that of
15 its counterpart. Reusability of the microfluidic device produced *via* 1-step method
16 was tested for three cycles of enzymatic reaction and at least 85% of the initial
17 productivity was maintained. The microfluidic device could be operated for up to 40
18 hours in a continuous flow and on average 70% of the initial productivity was
19 maintained. The simplified immobilisation method required fewer chemicals and less
20 time for preparation of the immobilised microfluidic device compared to the 3-step
21 method while achieving higher specific enzyme activity. The method represents a
22 promising approach for the development of immobilised enzymatic microfluidic
23 devices. Additionally, it also could be applied to immobilisation of other histidine-
24 tagged proteins (e.g. antibodies for immunosorbent assays).

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1 **Keywords**

- 2 Microfluidics; poly(methyl methacrylate) (PMMA), enzyme immobilisation;
- 3 histidine-tagged enzyme; transketolase.

1 **Introduction**

2 Microfluidic devices have found a wide range of applications in the fields of
3 biotechnology, chemistry and chemical engineering [1-6]. They offer many
4 advantages such as large surface to volume ratio in microchannels, enhanced mass
5 and heat transfer due to shorter diffusion paths [6], improved spatial and temporal
6 reaction control [7, 8].

7 The use of microfluidic devices has also increased in the context of enzymatic
8 applications. Several techniques, either with free or immobilised enzymes, that
9 demonstrate the potential of performing biocatalytic processes in microfluidic
10 systems have been reported [6,7,9,10]. For free enzymes in solution phase, there are
11 some bottlenecks restricting their use such as the need of **an** additional downstream
12 unit operation to recover and reuse them, and long-term operational stability [7]. In
13 contrast, use of immobilised enzymes offers advantages such as improved stability
14 and reusability; without the need for purification of the catalyst from substrates and
15 product [11-14]. However, immobilisation of enzymes on various surfaces can
16 sometimes affect biocatalyst performance.

17 Various materials such as glass [15-17], polystyrene [18], silicon [19,20],
18 underivatized silica [10,21] and poly(methyl methacrylate) (PMMA) [22], have
19 already been used to immobilise enzymes in microfluidic devices. Out of these,
20 PMMA is very favorable for rapid prototyping and is widely used for fabrication of
21 microfluidic devices, due to its low price, biocompatibility, excellent optical
22 transparency and attractive mechanical and chemical properties [23]. It has also
23 potential in **the** creation of fully disposable microfluidic devices favorable for **a** wide
24 range of applications, and ideal for preparing “green microchips” as it decomposes
25 into reusable monomer at a high temperature [23]. Developing fast and simple
26 immobilisation methods suitable for PMMA material are of fundamental importance
27 for **the** development of immobilised microfluidic systems.

28 A few approaches have been taken to immobilise enzymes in PMMA flow systems.
29 Cerqueira *et al.* covalently immobilised glucose oxidase enzyme in PMMA
30 microchannels using glutaraldehyde on a preliminarily aminated surface *via*

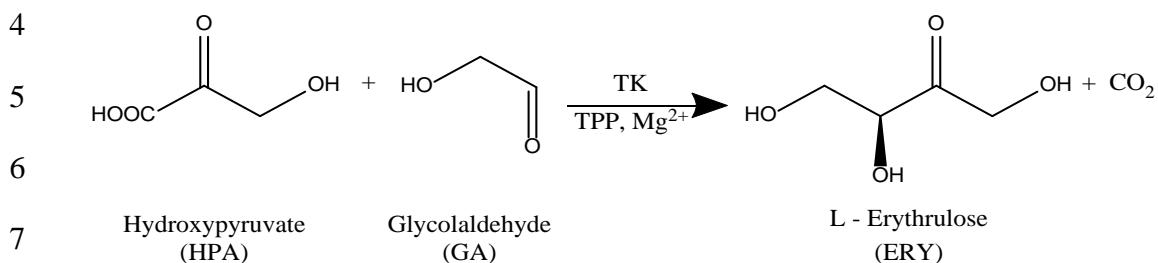
1 polyethyleneimine [24]. Llopis *et al.* reported a covalent immobilisation method
2 attaching active groups directly on PMMA surface of a microfluidic device using N-
3 lithiodiaminoethane, by exposure to UV radiation proceeded by N-(3-
4 dimethylaminopropyl)-N-ethylcarbodiimide coupling of ethylenediamine addition
5 [25]. Due to the inert nature of PMMA [26], the reported immobilisation methods
6 were based on PMMA surface functionalisation with amine groups or required
7 covalent binding of enzymes. The latter prevented reusability of the surface in
8 contrast to reversible binding methods such as the ones based on His-tag/Ni-NTA
9 interaction.

10 His-tag directed immobilisation method applied to PMMA plug flow bioreactor was
11 reported by Wollenberg *et al* [27]. The authors used coupling of CYP2C9 enzyme to
12 UV-activated PMMA *via* the Ni(II) chelator and histidine tag on the enzyme for
13 production of drug metabolites [27]. However, this method also contained surface
14 functionalisation steps making the immobilisation in the PMMA microfluidic device
15 more laborious and time consuming.

16 In this work, we report a simplified procedure for histidine-tagged enzyme
17 immobilisation on the surface of a PMMA microchannel using direct linking of N-
18 (5-Amino-1-carboxypentyl) iminodiacetic acid (AB-NTA) molecules to a modified
19 PMMA surface without the need for prior amination. Imaging and IR analysis of the
20 surface were carried out to investigate the proposed simplified immobilisation
21 method (from here onward called 1-step immobilisation method). We compared the
22 1-step immobilisation with an established immobilisation protocol (from here
23 onward called 3-step immobilisation method) that requires an amination step and use
24 of glutaraldehyde as a cross-linker prior to His-tag/Ni-NTA interaction.

25 Both techniques were applied to immobilise histidine-tagged transketolase (TK) in a
26 microfluidic device made out of PMMA. Transketolase enzyme was chosen due to
27 its wide substrate tolerance and high enantio- and regio-specificity [28], which make
28 it an attractive biocatalyst for the asymmetric synthesis of chiral metabolites [29-33]
29 and highly relevant in the pharmaceutical industry. However, the method is
30 applicable to a large variety of proteins that can be engineered with a His-tag [34].
31 The TK-catalysed conversion of hydroxypyruvate (HPA) and glycolaldehyde (GA)
32 for production of the chiral ketoalcohol L-erythrulose (ERY) was chosen as a model
33 reaction (Figure 1). We compared specific activities of TK obtained from continuous

1 reactions carried out in the microfluidic devices with TK immobilised using 1-step
2 and 3-step immobilisation methods. Reusability and operational stability studies of
3 the microfluidic device prepared by the 1-step immobilisation method were studied.



8 **Figure 1. Reaction scheme of a transketolase (TK) catalysed reaction.** Reaction scheme of the
9 synthesis of L-erythrulose by TK using hydroxypyruvate and glycolaldehyde as substrates. The use of
10 hydroxypyruvate as carbon donor enables the reaction to go to completion due to the release of carbon
11 dioxide formed as by-product. The transketolase was immobilised on the surface of a poly(methyl
12 methacrylate) (PMMA) device.

13 **Materials and Methods**

14 *Materials*

15 (5S)-N-(5-Amino-1-carboxypentyl)iminodiacetic acid (AB-NTA) and glutaraldehyde
16 were purchased from Insight Biotechnology (Wembley, UK). 1,6-Hexanediamine
17 (HMDA) was obtained from Fisher Scientific (Leicestershire, UK). SDS 4-20% Tris-
18 Glycine Mini-PROTEAN TGX™ Precast Gels and GFP recombinant *Aequorea*
19 *victoria* protein were purchased from BioRad Laboratories Ltd. (Hertfordshire, UK)
20 and Life Technologies (Paisley, UK), respectively. Poly(methyl methacrylate)
21 (PMMA) was obtained from RS (Corby, UK) while BSA stock solution and the BCA
22 protein assay kit were obtained from Thermo Fisher Scientific (UK). All other
23 reagents were purchased from Sigma-Aldrich (UK).

24 *Microfluidic device fabrication*

25 All components were designed using Adobe Illustrator CS6 software (Adobe
26 Systems Inc., USA). The device was comprised of two rigid 2 mm PMMA layers.
27 One of the layers has a channel with additional ridges for increasing the surface area
28 to volume ratio and the total length is 280 mm (Figure S1). The channels and cutouts

1 were fabricated using a CO₂ laser marking head (Zing, USA) and the layers were
2 thermally bonded (2 h, 102.5 °C). Interconnect ports were laser cut from 6 mm thick
3 PMMA, with two holes tapped with a M3 thread for attachment of the connector to
4 the device, and an M6 threaded hole to allow standard connection tubing (P-221,
5 Upchurch Scientific, Oak Harbor, USA) to be attached.

6 To determine the experimental volume of the microfluidic device, the device with
7 clean and dry microchannel was connected to a syringe pump (AL 1000-220). Water
8 was pumped at a fixed flow rate and time that it takes for the water to flow from the
9 inlet to the outlet was measured. Reactor volume was calculated by multiplying the
10 given flow rate by the measured time.

11 ***Biocatalyst preparation***

12 Transketolase stocks of *E. coli* BL21gold (DE3) with plasmid pQR791 (His₆-TK)
13 were produced in-house and stored at -80 °C in LB broth containing 50% (v/v)
14 glycerol. In order to obtain higher enzyme yields, MagicMedia *E.coli* expression
15 medium (ThermoFisher Scientific, UK) was used. The complete MagicMedia
16 medium was prepared using the supplied kit components according to its manual.
17 The protocol for biocatalyst preparation was similar to that previously described
18 [35]. Briefly, overnight cultures of *E.coli* were prepared in flasks containing 10 ml of
19 LB broth with 150 µg·mL⁻¹ ampicillin. Cells were sub-cultured using 1% (v/v)
20 inoculum in 1 L baffled flasks containing 200 mL of the prepared complete
21 MagicMedia medium with 150 µg·mL⁻¹ ampicillin and incubated at 37 °C and 250
22 rpm until the bacterial growth reached stationary phase. Cells were harvested by
23 centrifugation at 4000 rpm for 20 minutes. The cell pellets were resuspended in 50
24 mM Tris-HCl pH 7.0 and sonicated on ice (Soniprep 150, MSE Sanyo, Japan). The
25 suspension was centrifuged at 13,000 rpm and 4 °C for 20 min. The supernatant was
26 filtered using PVDF syringe filters (Whatman, Maidstone, UK) with a molecular
27 weight cutoff of 0.2 µm and stored at -20 °C until further use.

1 ***Preparation of poly(methyl methacrylate) microfluidic device with immobilised***
2 ***His-tagged transketolase***

3 ○ *3-step immobilisation protocol*

4 Poly(methyl methacrylate) (PMMA) microfluidic device channel was functionalised
5 and immobilised with His-tagged enzyme according to protocols adapted from Fixe
6 *et al.* and Oshige *et al.* [34, 36]. Briefly, a channel in the PMMA microfluidic device
7 was flushed and filled with isopropanol (99%) and incubated at 30 °C for 3 hours.
8 The microfluidic device channel was then rinsed thoroughly with Milli-Q water and
9 incubated with 10% (v/v) hexamethylene-diamine (HMDA) in 100 mM borate buffer
10 pH 11.5, for 2 h. The channel was then thoroughly flushed with Milli-Q water for
11 several channel volumes. Afterwards, the channel was incubated with 1% (v/v)
12 glutaraldehyde overnight at 37 °C. Another overnight incubation at 37 °C was done
13 with a 0.05% (w/v) solution of N-(5-amino-1-carboxy-pentyl) iminodiacetic acid
14 (AB-NTA) in 0.1 M HEPES buffer, pH 8.0. Then, the channel was flushed with
15 Milli-Q water using syringe pump AL 1000-220 (World Precision Instruments,
16 USA) at flow rates of 20 $\mu\text{l}\cdot\text{min}^{-1}$. Finally, a solution of 0.5 M NiCl_2 was pumped
17 through the microchannel at a flow rate of 10 $\mu\text{l}\cdot\text{min}^{-1}$ for 1 hour followed by a Milli-
18 Q wash.

19 ○ *1-step immobilisation protocol*

20 1-step immobilisation protocol was adapted from the 3-step immobilisation by direct
21 conjugation of AB-NTA molecules to available methyl ester bonds on PMMA
22 surface. For this purpose, the channel was filled with isopropanol (99%) and
23 incubated at 30°C for 3 hours. After rinsing with Milli-Q water, the channel was
24 incubated overnight with 0.05% (w/v) AB-NTA in 0.1 M HEPES buffer, pH 8.0.
25 Analogously to the 3-step immobilisation, the channel was rinsed with Milli-Q water
26 and then a 0.5 M NiCl_2 solution was pumped through the microchannel at a flow rate
27 of 10 $\mu\text{l}\cdot\text{min}^{-1}$ for 1 hour. The channel was washed with Milli-Q before TK
28 immobilisation.

29 ***Transketolase immobilisation and elution***

30 Cell lysates containing TK ranging from 20% to 28% (w/w) were loaded at a protein
31 concentration of $14.8\pm 2.6 \text{ mg}\cdot\text{ml}^{-1}$ and $14.3\pm 3.1 \text{ mg}\cdot\text{ml}^{-1}$ for the 1-step and 3-step

1 immobilised microfluidic devices, respectively. The lysates were pumped into the
2 microchannels using a syringe pump AL 1000-220 (World Precision Instruments,
3 USA) at a flow rate of $5 \mu\text{l}\cdot\text{min}^{-1}$ at 4°C for several channel volumes. After 1 h a
4 solution of 50 mM Tris-HCL, pH 7.5 was pumped through the microchannels at 20
5 $\mu\text{l}\cdot\text{min}^{-1}$ to remove non-specifically bound enzyme. Samples were collected
6 periodically and assayed for protein content. After operational stability studies, the
7 bound enzyme was removed by treating the channel with EDTA elution buffer (50
8 mM EDTA; pH 8.0) at $20 \mu\text{l}\cdot\text{min}^{-1}$ for at least 2 channel volumes. Collected samples
9 were concentrated down to 75 μl volume using Amicon Ultra Centrifugal filters
10 (30,000 NMWL).

11 *Characterisation of immobilisation surface*

12 For Scanning Electron Microscopy (SEM) imaging studies, three PMMA devices
13 were fabricated following the procedure as described earlier. One microchannel
14 surface was used as a control and remained untreated. The second and third
15 microfluidic device channel surfaces were modified using the 1-step immobilisation
16 protocol and washed to remove any excess or non-specific binding. After this, the
17 third microfluidic channel surface was used for TK immobilisation as described
18 earlier and washed with Milli-Q to remove any non-specifically bound enzyme. The
19 channel surfaces were dried and sputter-coated by a mix of gold and platinum
20 nanoparticle layer to allow good conductivity on the surface. SEM imaging of the
21 pre-treated samples was carried out using JEOL 5610LV system with magnifications
22 ranging from 100 to 20 000-fold.

23 Fourier transform infrared (FT-IR) spectra of bare PMMA and the AB-NTA
24 conjugated PMMA surfaces were recorded on Bruker platinum ATR. Samples were
25 two microchannel surfaces that were laser cut in 2mm PMMA slabs following the
26 fabrication procedure described earlier. Bare PMMA surface was cleaned and dried
27 before measurement. Following 1-step immobilisation procedure on a bare PMMA
28 surface and after incubation with AB-NTA, the surface was washed with Milli-Q to
29 remove any non-bound AB-NTA molecules. The surface was dried before FT-IR
30 spectroscopy.

1 ***Transketolase reactions in microfluidic device***

2 Transketolase reaction in the microfluidic device was carried out at 4 °C. A cofactor
3 solution of 4.8 mM thiamine diphosphate (ThDP) and 19.6 mM magnesium
4 dichloride (MgCl₂) in 50mM Tris-HCL, pH 7.0 was pumped through the
5 microchannel at 10 μl·min⁻¹ (KDS100, KD Scientific, Holliston, USA) for 30 min.
6 Afterwards, a substrate mix of 12.5 mM glycolaldehyde (GA) and 12.5 mM
7 hydroxypyruvate (HPA) in 50mM Tris-HCL pH 7.0 was flown through the reactor at
8 flow rates ranging from 2.3 to 30 μl·min⁻¹ (mean residence times of 25 to 1 min,
9 respectively). Samples generated from each flow rate were collected into Eppendorf
10 tubes containing 0.1% (v/v) trifluoroacetic acid and analysed by HPLC.

11 ***Reusability and operational enzyme stability studies***

12 After each reaction run, the microfluidic device was washed with 50 mM EDTA pH
13 8.0 at 20 μl·min⁻¹ for two microchannel volumes at room temperature. Samples were
14 collected and assayed for protein content. The microfluidic device was stored at -20
15 °C until further use. For the next cycle of immobilisation, the microfluidic device
16 was first washed with Milli-Q water at 20 μl·min⁻¹ and then the surface was
17 regenerated with NiCl₂ as described in 1-step immobilisation method section.
18 Loading of TK was carried out as mentioned in transketolase immobilisation and
19 elution section.

20 Operational stability study was conducted in a microfluidic device with TK
21 immobilised through 1-step immobilisation. After TK immobilisation and washing
22 steps, TK reaction was carried out under a constant flow rate of 5 μl·min⁻¹ for up to
23 40 hours. Samples were collected at various times into tubes with 0.1% (v/v)
24 trifluoroacetic acid and analysed for ERY production by HPLC.

25 ***Analytcs***

26 ○ *SDS-PAGE and protein concentration determination*

27 SDS-PAGE protein analysis was performed using a Mini-155 Protean II system
28 (Bio-Rad Laboratories Inc., Hemel Hempstead, UK) with SDS 12% Tris-Glycine
29 (BioRad, UK) pre-cast gels using 10x Tris/Glycine/SDS electrophoresis buffer
30 (BioRad, UK). Gels were stained with Instant Blue (Expedeon Ltd., Cambridge,

1 UK). Destaining was performed in Milli-Q water overnight and a *Gel-Doc-it*
2 bioimaging system (Bioimaging systems, Cambridge) was used to visualize the gels.
3 Protein concentration was determined in triplicates using the Micro BCA protein
4 assay kit (Thermo Scientific, UK) according to the manufacturer instructions.
5 Absorbance measurements at 561 nm were carried out on an ATI Unicam UV/VIS
6 spectrophotometer (Spectronic, Leeds, UK). Protein concentration was quantified
7 using a calibration curve with bovine serum albumin as standard.

8 ○ *TK mass and specific activities of transketolase*

9 TK amount of eluted enzyme was determined by SDS-PAGE densitometry.
10 Densitometry of samples electrophoresed on a 12% SDS-PAGE gel (Bio-Rad) was
11 used. To calculate the mass of eluted enzyme in elution studies, imaging and
12 quantification was done using GE Amersham Imager 600. A range of concentrations
13 of commercial BSA was run on an SDS-Page gel to obtain a standard curve based on
14 integrated band density for calibration of the enzyme concentration. BSA
15 concentration of 0.5 mg·mL⁻¹ was used as a reference band on all other gels to
16 account for variation in density obtained from different gels.

17 For enzyme binding efficiency and consequent specific activity determination studies
18 previously imaged gels were analysed using ImageQuant TL Version 8.1 to calculate
19 the TK band density relative to the total band density. Total protein in the samples
20 was determined using a BCA assay.

21 ***Substrate and product quantification of TK-catalysed reactions***

22 L-erythrulose (L-ERY) and hydroxypyruvate (HPA) were quantified with HPLC
23 (Ultimate 3000, Thermo, UK) using an Aminex HPX-87 column (300 mm x 7.8 mm,
24 Bio-Rad, UK) at 60 °C using an isocratic flow of 0.1% (v/v) aqueous trifluoroacetic
25 acid at 0.6 mL·min⁻¹. Compounds were detected at 210 nm.

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1 **Results and discussion**

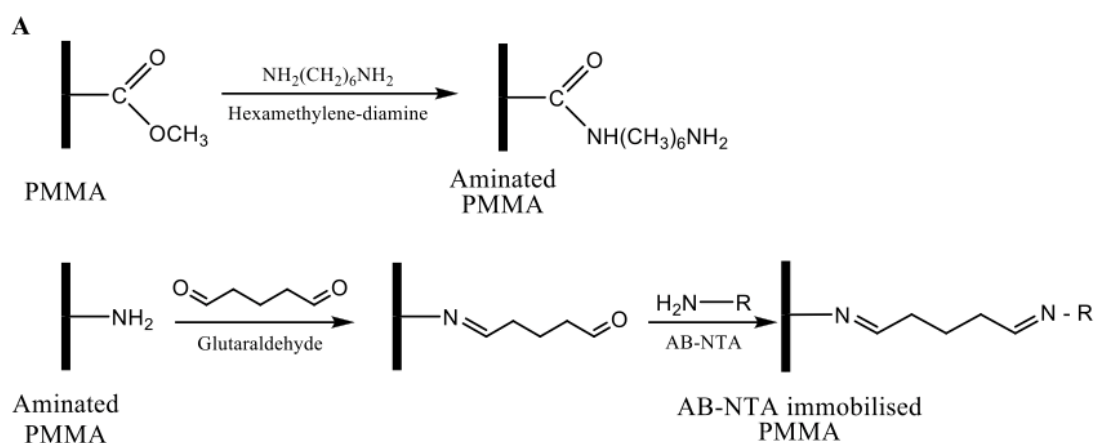
2 *Immobilisation of transketolase on to poly(methyl methacrylate) surface*

3 Many protocols for immobilising enzymes on to polymeric materials require prior
4 surface modification, since the material surface does not contain suitable functional
5 groups in their native form [37]. For example, the protocol for immobilisation of
6 His-tagged proteins on various polymers developed by Oshige *et al.* requires surface
7 amination in order to bind glutaraldehyde that is used as a cross-linker enabling
8 enzyme binding *via* Ni-NTA [34].

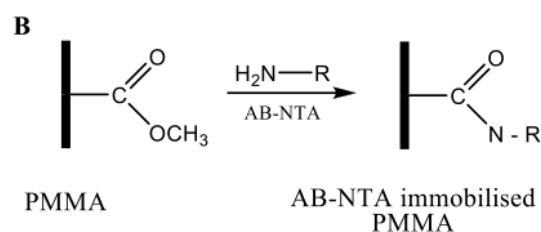
9 In this work, a 3-step immobilisation method adapted from Oshige *et al.* based on
10 amination of PMMA was investigated first [34]. The protocol included amination of
11 the PMMA surface by using hexamethylene-diamine (HMDA). Glutaraldehyde was
12 used as a cross-linker to conjugate amine bonds of N-(5-Amino-1-carboxy-pentyl)
13 iminodiacetic acid (AB-NTA) to the aminated PMMA surface as shown in Figure
14 2A. However, application of this method was time-consuming, since it took two
15 overnight incubations to complete the surface treatment.

16 To establish a simpler protocol with concomitant reduction of surface modification
17 steps and the time involved, we developed an alternative method where AB-NTA
18 binds to available methyl esters of PMMA under basic pH conditions forming amide
19 bonds as shown in Figure 2B. Conjugating AB-NTA on to the surface of PMMA
20 eliminates the amination step required in existing protocols for enzyme
21 immobilisation on PMMA and the need of using glutaraldehyde required in the 3-
22 step immobilisation method.

3-step immobilisation method



1-step immobilisation method



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2 **Figure 2. 1-step and 3-step immobilisation chemistries on poly(methyl methacrylate) (PMMA)**
3 **surface.** (A) 3-step immobilisation chemistry. The first line corresponds to amination chemistry of the
4 surface. The available methyl esters of PMMA, under basic pH conditions, are reacted with an
5 electron donor (N) present on the hexamethylene-diamine (HMDA), producing primary amines on the
6 surface. The second line of the scheme represents the linking step of primary amine bonds formed on
7 PMMA surface with amine bonds of AB-NTA molecule using the cross-linker glutaraldehyde. (B) 1-
8 step immobilisation chemistry. AB-NTA molecule substitutes the HMDA step corresponding to 3-
9 step immobilisation procedure, and amine bonds of AB-NTA molecule react with the available methyl
10 esters on the PMMA surface, under basic pH conditions. These 1-step and 3-step immobilisation
11 procedures produce a functionalised PMMA surface that subsequently is used for immobilising
12 histidine-tagged enzymes.

13 *Characterisation of PMMA microchannel surfaces using scanning electron* 14 *microscopy and infrared spectroscopy*

15 To confirm the presence of AB-NTA and consequent immobilisation of TK using 1-
16 step immobilisation method, modified PMMA surfaces were investigated using
17 infrared spectroscopy and scanning electron microscopy. Figure 3 shows Fourier
18 transform infrared (FT-IR) spectra and respective scanning electron microscopy
19 images of unmodified PMMA channel surface (A), AB-NTA treated channel surface

1 (B), and of modified PMMA surface with TK immobilised *via* 1-step immobilisation
2 method (without IR spectra) (C, D).

3 The surface of open microchannel before modification steps showed roughness and
4 re-settled material particles and small cavities due to the laser cutting
5 microfabrication process (Figure 3A SEM image). Roughness is very favorable for
6 our application, as it potentially increases the effective area to be modified. The FT-
7 IR spectrum of unmodified PMMA channel shown in Figure 3A presented a set of
8 bands that are characteristic of pure PMMA surfaces [38].

9 The coupling of AB-NTA to the PMMA surface under basic pH conditions was
10 confirmed by the FT-IR spectrum (Figure 3B), where the peak at $\sim 3400\text{ cm}^{-1}$ can be
11 assigned to the N-H stretching vibrations due to amide bond formation [36]. As can
12 be noticed in respective SEM image of the channel, the presence of AB-NTA on
13 inner walls does not lead to significant changes in surface profile (Figure 3B SEM
14 image).

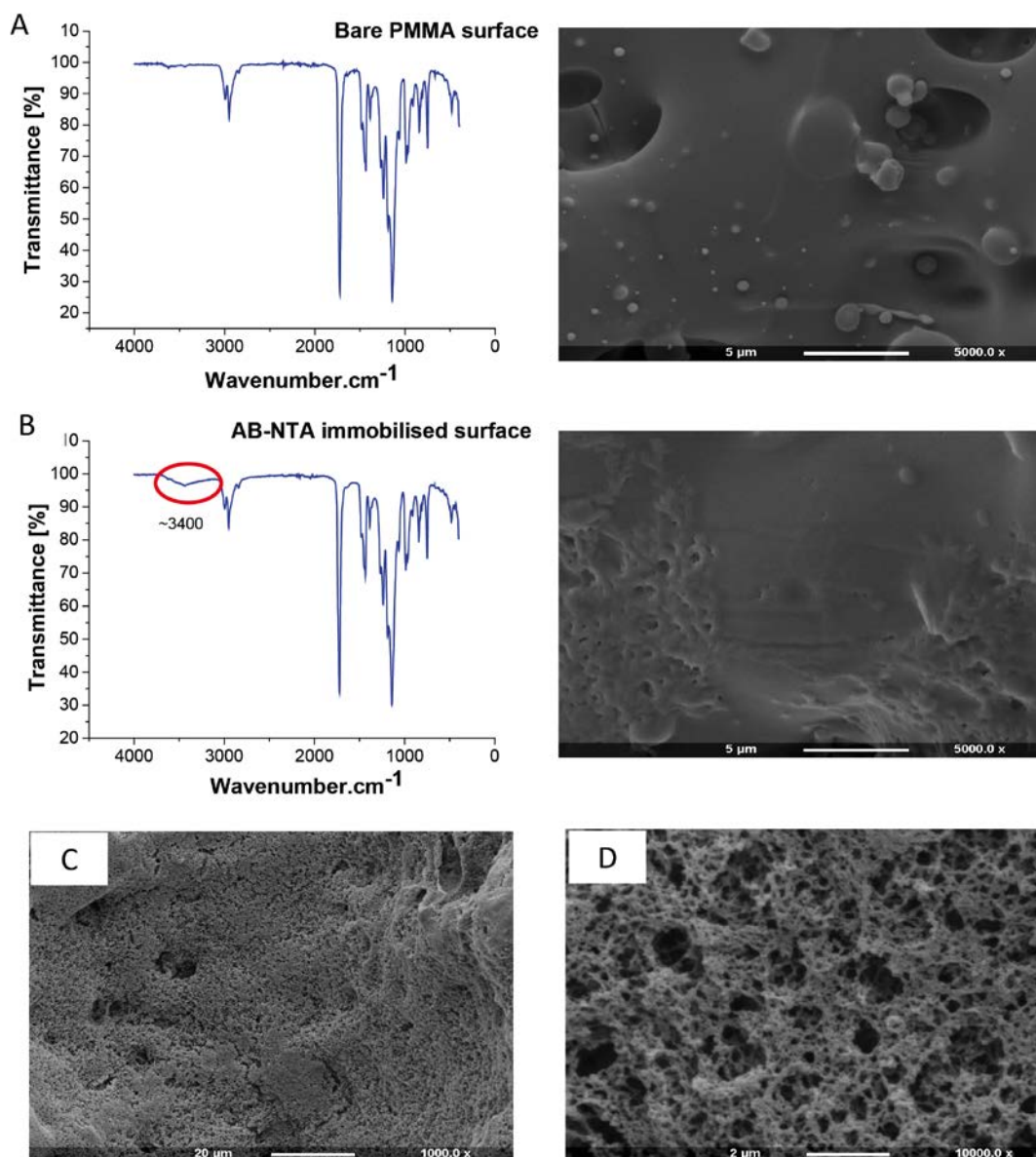
15 Figure 3(C) and (D) show the SEM images of surfaces with TK immobilised *via* the
16 1-step method on to PMMA microchannel surface. These images presented porous
17 structure that was distinctly different from the AB-NTA treated surface (B)
18 suggesting that these were enzyme structures bound on to the surface. We presume
19 that due to the apparent porous surface of the microchannel, the attached enzyme
20 formed an uneven layer or multiple layers.

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2 **Figure 3. Set of Fourier transform infrared spectroscopy spectra and respective scanning**
 3 **electron microscopy (SEM) images of (A) bare poly(methyl methacrylate (PMMA) microchannel**
 4 **surface; (B) microchannel surface treated with AB-NTA according to 1-step immobilisation method**
 5 **(C) 1000x and (D) 10,000x magnifications of PMMA channel surface after immobilisation of**
 6 **transketolase using 1-step immobilisation method.**

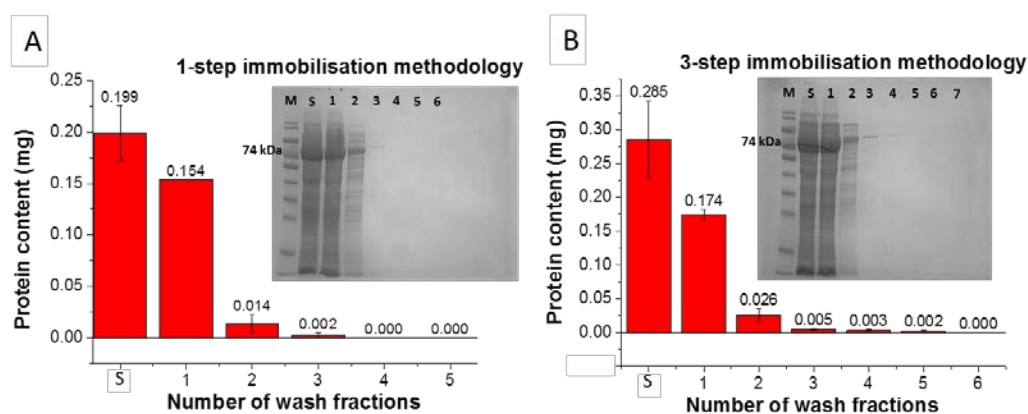
7 ***Enzyme binding efficiency in immobilised microfluidic devices***

8 A microfluidic device fabricated out of poly(methyl methacrylate) with a
 9 microchannel and additional grooves for **the** enhanced surface area was used. The

1 internal surface area of the microfluidic channel was calculated as 1,040 mm²
2 (Supplementary Materials, Section S1). The calculated volume was 80 μL assuming
3 sharp edges of the designed grooves inside the microchannel. However, the
4 experimentally determined volume was only 56 μL. The difference is likely due to
5 the non-rectangular profiles of the microchannel surface, which can be seen from
6 Figure S2 and therefore, experimentally determined volume was used in all
7 calculations in this work.

8 Immobilisation of TK was carried out using a cell lysate obtained from an *E. coli*
9 shake flask culture overexpressing a wild-type TK enzyme. Clarified cell lysates at
10 concentrations of 14.76 and 14.31 mg·ml⁻¹ (containing 20% and 28% TK
11 respectively determined *via* densitometry) were used for loading into the
12 microfluidic devices modified by 1-step- or 3-step immobilisation methods,
13 respectively. The enzyme amount retained in the microchannel was determined by a
14 mass balance between the amount of enzyme loaded at the inlet and that measured at
15 the outlet and in the wash fractions. In total, 6 to 7 washing steps were performed
16 with each collected sample containing a solution of one channel volume with the
17 outlet tubing volume (112 μL for 1-step and 124 μL for 3-step immobilised
18 microfluidic devices, respectively). Protein quantification of wash fractions after
19 enzyme immobilisation is shown in Figures 4A and 4B. Using SDS-PAGE
20 densitometric analysis, percentages of TK relative to total protein in each lane of
21 SDS-PAGE gels depicted in the insets of Figures 4A and 4B were determined and
22 used for calculating TK content in each wash fraction. No protein was detected after
23 the third wash of the microchannel immobilised with TK *via* 1-step immobilisation
24 method (Figure 4A inset), while the 3-step microfluidic device samples showed that
25 small amounts of protein were still detected up to the fifth wash fraction (Figure 4B
26 inset). To note, a control study of TK loading into bare non-modified PMMA
27 microfluidic device showed protein wash outs for the entire 7 washes and no
28 retention of the enzyme in the channel was observed as expected (Figure S3). Based
29 on the mass balance, the amount of bound TK in the microchannels was estimated to
30 be 29±15 μg for 1-step and 75±37 μg for 3-step immobilised TK microfluidic
31 devices, respectively. The amounts represented an average binding efficiency of
32 ~15% for the 1-step and ~26% for the 3-step methods and suggested that
33 approximately 6 and 15 times more enzyme was immobilised in the microfluidic

1 device, respectively, than the theoretically estimated amount for a microchannel of
 2 such geometry. The theoretical amount was estimated assuming monolayer
 3 immobilisation of TK in the microchannel and microfluidic device surface was found
 4 to accommodate approximately 5 μg of transketolase. Details of calculations are
 5 shown in Supplementary Materials, Section S1. The finding supports our
 6 interpretation of SEM images that the porous structure formed increases the available
 7 area for enzyme binding significantly. It is also in agreement with Miyazaki *et al.*
 8 [39] who reported 8-10 fold higher enzyme amount relative to monolayer coverage
 9 for a His-tagged L-lactic dehydrogenase enzyme immobilised in a derivatised silica
 10 microchannel. Additionally, elution of TK that was immobilised in the microchannel
 11 using 1-step immobilisation method produced a faint band corresponding to TK
 12 molecular weight (Figure S4). Densitometric analysis of the band showed that
 13 approximately 9 μg of TK was recovered from the elution of the microfluidic device
 14 prepared using the 1-step method, which comprised $\sim 31\%$ of the average amount of
 15 TK calculated from the mass balance. This eluted amount of TK was almost twice
 16 higher than the theoretical value calculated from the geometry of the channel. The
 17 difference between the bound TK calculated from the mass balance and that obtained
 18 from the eluted band is likely due to the difficulty of accurately quantifying such
 19 small amounts of enzyme.



20

21 **Figure 4. Quantification of transketolase (TK) protein concentration to determine enzyme**
 22 **binding efficiency of immobilised microfluidic devices prepared by 1-step and 3-step**
 23 **immobilisation methods.** Protein amounts loaded into the microfluidic device and collected in flow-
 24 through (wash) fractions from the microchannels with immobilised TK *via* (A) 1-step and (B) 3-step
 25 methods. Inset: SDS-PAGE gel of protein samples. Lane M corresponds to SDS marker with the 74
 26 kDa band indicative of the size of the TK monomer (72.5 kDa), Lane S shows the amount of protein
 27 loaded (S= $\sim 0.20 \pm 0.03$ mg TK for 1-step and 0.29 ± 0.06 mg TK for 3-step immobilised microfluidic

1 devices), and Lanes 1-7 represent the protein content in collected wash fractions. Final immobilised
2 amount of TK enzyme in microfluidic devices using 1-step and 3-step methods was calculated to be
3 $29 \pm 15 \mu\text{g}$ and $75 \pm 36 \mu\text{g}$, respectively. Protein detection was done in triplicates. Error bars represent
4 standard deviation above the mean ($n=3$).

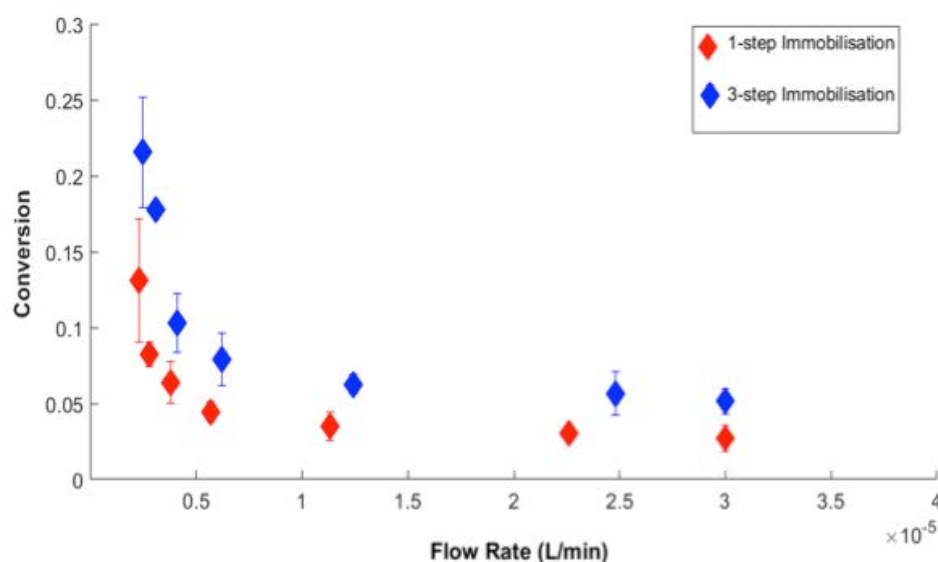
5 ***Transketolase reactions in continuous modes***

6 In order to compare specific activities of TK in two PMMA microfluidic devices
7 immobilised through 1-step and 3-step methods, the L-erythrulose production was
8 investigated. Previously, the transketolase-catalysed reaction has been characterised
9 in continuous flow microreactors with the enzyme immobilised either on the
10 microchannel walls or on Ni-NTA agarose beads [10, 21]. By contrast, in this work,
11 a different immobilisation method suitable for PMMA material was used and the
12 device was operated at higher flow rates ranging from 2 to $\sim 30 \mu\text{l}\cdot\text{min}^{-1}$.
13 Additionally, to ensure the prolonged stability of the enzyme over extended periods
14 of time the continuous flow reaction was run at $4 \text{ }^\circ\text{C}$ instead of room temperature.
15 This would minimise possible thermal deactivation of the enzyme.

16 Dependence of enzyme conversion on flow rate shown in Figure 5 yielded similar
17 profiles for microfluidic devices immobilised with 1-step or 3-step methods. Specific
18 activities of TK in the two devices were found to be 124 ± 13 and $88 \pm 11 \text{ nmol}\cdot\text{mg}^{-1}$
19 $\cdot\text{min}^{-1}$ for 1-step and 3-step immobilisation, respectively. Data were derived from
20 the product formation determined as a function of residence time shown in Figure S5
21 in Supplementary Materials section. For immobilised TK enzyme on Ni-NTA
22 agarose beads previously Halim *et al.* [21] reported a specific activity of 8.20
23 $\mu\text{mol}\cdot\text{mg}^{-1}\cdot\text{min}^{-1}$, while Matosevic *et al.* [10] reported a specific enzyme activity of
24 $0.8 \mu\text{mol}\cdot\text{mg}^{-1}\cdot\text{min}^{-1}$ for TK immobilised on the walls of a silica capillary. The
25 reduced specific activity found in this study can be explained by the lower
26 temperature ($4 \text{ }^\circ\text{C}$) used in comparison with previous studies that were carried out at
27 room temperature.

28 Additionally, the immobilised TK retained only $\sim 4\%$ of the specific activity of the
29 free enzyme in solution that was found to be $3.38 \mu\text{mol}\cdot\text{mg}^{-1}\cdot\text{min}^{-1}$ (Figure S6). This
30 is in line with earlier data by Matosevic [40], who reported also 4% of the free
31 enzyme activity was retained using His-tagged immobilisation. For covalent
32 attachment via cross-linking with glutaraldehyde onto the amino-silanized surface

1 Thomsen *et al.* reported that immobilised β -glycoside hydrolase CelB in
2 microreactor retained only about 3% of the specific activity of the free enzyme,
3 while the CelB attached to silanized macroporous glass beads in batch, that
4 employed the same immobilisation technique, retained 35% of the free enzyme
5 specific activity [41]. Such large differences between immobilised and free enzyme
6 specific activities could be caused by mass transfer limitations due to an unfavorable
7 conformation of the immobilised enzyme and/or multi-layer attachment and possibly
8 some inactivation of the enzyme in an immobilised form [38].



9

10 **Figure 5. Conversion of glycolaldehyde (GA) and hydroxypyruvate (HPA) to L-erythrulose (L-**
11 **ERY) at 4 °C as a function of flow rate in the poly(methyl methacrylate) (PMMA) microfluidic**
12 **devices.** Comparison of product formation obtained from microfluidic devices with transketolase
13 (TK) immobilised *via* 1-step and 3-step immobilisation techniques. The specific activity values
14 derived from this data were 124 ± 13 and 88 ± 11 $\text{nmol} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ for 1-step and 3-step immobilisation,
15 respectively. Error bars represent one standard deviation above the mean ($n=2$). TK-catalysed
16 reactions were performed using equimolar concentrations of GA and HPA (12.5 mM) each.

17 ***Reusability of immobilised microfluidic device prepared by 1-step method***

18 To test reusability of the 1-step immobilisation layer, the performance of the
19 microfluidic device was tested over multiple immobilisation-elution-regeneration
20 cycles. Previously reported immobilised microreactors have shown better retention
21 of enzyme activity than using enzymes in free solutions. For instance, Thomsen *et al.*
22 demonstrated consistent productivity of a β -glycoside hydrolase CelB-immobilised
23 microreactor over 5 days [38]. Matosevic *et al.* showed successful reusability of a

1 silica-based immobilised microreactor over 5 cycles [10]. In this work, the suitability
2 of 1-step immobilisation protocol for multiple usages over longer periods of time
3 was investigated. The reusability of the microfluidic device with TK immobilised *via*
4 1-step method was studied by using the same microchannel after eluting the enzyme
5 and reusing the microreactor in three successive runs each with a fresh enzyme. High
6 productivity levels close to 100% were maintained over 3 cycles operated at a flow
7 rate of 10 $\mu\text{l}\cdot\text{min}^{-1}$, although a reduction to 85% was observed in the second reuse
8 (Figure S7A). It can be concluded the immobilisation surface was stable and activity
9 of TK was maintained for at least 21 days.

10 In addition, the operational stability of the TK immobilised microfluidic device *via*
11 1-step immobilisation method was investigated under continuous flow over 40 hours
12 at a flow rate of 5 $\mu\text{l}\cdot\text{min}^{-1}$. Samples were taken after 1, 14 and 39 hours to measure
13 the conversion of HPA and GA to ERY product. The productivity of the TK
14 immobilised in the device decreased to around 65% after 14 h but on average stayed
15 around 70% for the duration of the study (Figure S7B). The results are similar to
16 previously obtained values of operational stability of TK immobilised in a packed-
17 bed reactor, where productivity dropped to 76% after 48 h of reaction time [19].
18 However, it is worth noting that flow rates used in this study were 5 times higher
19 than the flow rates used in the reported study (1 $\mu\text{l}\cdot\text{min}^{-1}$), which may explain the
20 slight difference in productivities of the immobilised TK.

21 **Conclusions**

22 In this work, 1-step immobilisation of a His-tagged enzyme was investigated as a fast
23 and simple alternative to existing methods for immobilising enzymes in PMMA
24 microfluidic devices. The 1-step method was compared to the more common 3-step
25 immobilisation of histidine-tagged enzymes in PMMA microchannels. We assessed
26 the chemistries of the two immobilisation techniques and corresponding surface
27 preparation steps, their enzyme binding efficiencies in microfluidic devices as well
28 as the specific activities of the immobilised enzyme and conversion yields (Table 1).
29 The 1-step immobilisation method holds potential advantages over other protocols.
30 One of the advantages of the 1-step method is the much shorter preparation time.

1 Additionally, the number of chemicals required for 1-step immobilisation is
2 significantly less, since this method does not require a separate surface amination of
3 PMMA, unlike the 3-step immobilisation protocol. As a result, the immobilisation
4 cost may potentially be reduced. Furthermore, the presented method requires fewer
5 wash steps. The simplified 1-step protocol yielded approximately 10% lower enzyme
6 binding efficiency than the 3-step immobilisation method, however, produced similar
7 L-erythrulose conversions as a function of flow rate due to higher specific enzyme
8 activity. The 1-step method presents a viable approach for the development of
9 enzymatic microfluidic devices and could potentially be applied to combine enzyme
10 purification with immobilisation of His-tagged proteins from crude cell extracts.
11 Additionally, it also could be applied to immobilisation of other histidine-tagged
12 proteins (e.g. antibodies for immunosorbent assays).

13

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10 Microscopy, FT-IR spectroscopy, and biocatalyst preparation, respectively.

11 **Author Contributions**

12 G.K. and F. B. conceived, designed and G.K. performed the experiments; G.K. N.D.,
13 M.M. and F.B. analysed the data; G.K. wrote the manuscript with input from F.B.,
14 N.D. and M.M.; N.S and F.B. edited the paper.

15 **Conflicts of Interest**

16 The authors declare no conflict of interest.

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35

1 **Table 1. Comparison of key characteristics of the 1-step and 3-step immobilisation of His-tagged**
2 **transketolase in poly(methyl methacrylate) PMMA microfluidic devices.**

3

Characteristic	1-step immobilisation	3-step immobilisation
Preparation time	1 day	2-3 days
Number of chemicals required	3	7
Binding efficiency, (%)	~15	~26
Specific activity, (nmol·mg ⁻¹ ·min ⁻¹)	124±13	88±11

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1 **Figure Captions**

2 **Figure 1. Reaction scheme of a transketolase (TK) catalysed reaction.** Reaction
3 scheme of the synthesis of L-erythrulose by TK using hydroxypyruvate and
4 glycolaldehyde as substrates. The use of hydroxypyruvate as carbon donor enables
5 the reaction to go to completion due to the release of carbon dioxide formed as by-
6 product. The transketolase was immobilised on the surface of a poly(methyl
7 methacrylate) (PMMA) device.

8

9 **Figure 2. 1-step and 3-step immobilisation chemistries on poly(methyl**
10 **methacrylate) (PMMA) surface.** (A) 3-step immobilisation chemistry. The first line
11 corresponds to amination chemistry of the surface. The available methyl esters of
12 PMMA, under basic pH conditions, are reacted with an electron donor (N) present on
13 the hexamethylene-diamine (HMDA), producing primary amines on the surface. **The**
14 second line of the scheme represents the linking step of primary amine bonds formed
15 on PMMA surface with amine bonds of AB-NTA molecule using the cross-linker
16 glutaraldehyde. (B) 1-step immobilisation chemistry. AB-NTA molecule substitutes
17 the HMDA step corresponding to 3-step immobilisation procedure, and amine bonds
18 of AB-NTA molecule react with the available methyl esters on the PMMA surface,
19 under basic pH conditions. These 1-step and 3-step immobilisation procedures
20 produce a functionalised PMMA surface that subsequently is used for immobilising
21 histidine-tagged enzymes.

22

23 **Figure 3. Set of Fourier transform infrared spectroscopy spectra and respective**
24 **scanning electron microscopy (SEM) images of** (A) bare poly(methyl)
25 methacrylate (PMMA) microchannel surface; (B) microchannel surface treated with
26 AB-NTA according to 1-step immobilisation method (C) 1000x and (D) 10,000x
27 magnifications of PMMA channel surface after immobilisation of transketolase using
28 1-step immobilisation method.

29

1 **Figure 4. Quantification of transketolase (TK) protein concentration to**
2 **determine enzyme binding efficiency of immobilised microfluidic devices**
3 **prepared by 1-step and 3-step immobilisation methods.** Protein amounts loaded
4 into the microfluidic device and collected in flow-through (wash) fractions from the
5 microchannels with immobilised TK *via* (A) 1-step and (B) 3-step methods. Inset:
6 SDS-PAGE gel of protein samples. Lane M corresponds to SDS marker with the 74
7 kDa band indicative of the size of the TK monomer (72.5 kDa), Lane S shows the
8 amount of protein loaded ($S \sim 0.20 \pm 0.03$ mg TK for 1-step and 0.29 ± 0.06 mg TK for
9 3-step immobilised microfluidic devices), and Lanes 1-7 represent the protein
10 content in collected wash fractions. Final immobilised amount of TK enzyme in
11 microfluidic devices using 1-step and 3-step methods was calculated to be 29 ± 15 μ g
12 and 75 ± 36 μ g, respectively. Protein detection was done in triplicates. Error bars
13 represent standard deviation above the mean ($n=3$).

14

15 **Figure 5. Conversion of glycolaldehyde (GA) and hydroxypyruvate (HPA) to L-**
16 **erythrulose (L-ERY) at 4 °C as a function of flow rate in the poly(methyl**
17 **methacrylate) (PMMA) microfluidic devices.** Comparison of product formation
18 obtained from microfluidic devices with transketolase (TK) immobilised *via* 1-step
19 and 3-step immobilisation techniques. The specific activity values derived from this
20 data were 124 ± 13 and 88 ± 11 $\text{nmol} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ for 1-step and 3-step immobilisation,
21 respectively. Error bars represent one standard deviation above the mean ($n=2$). TK-
22 catalysed reactions were performed using equimolar concentrations of GA and HPA
23 (12.5 mM) each.

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