



Natural Products Synthesis Hot Paper

International Edition: DOI: 10.1002/anie.201712065 German Edition: DOI: 10.1002/ange.201712065

Enantioselective Synthesis of the Cyclopiazonic Acid Family Using **Sulfur Ylides**

Oleksandr Zhurakovskyi, Yunus E. Türkmen, Lorenz E. Löffler, Vijayalakshmi A. Moorthie, C. Chun Chen, Michael A. Shaw, Mark R. Crimmin, Marco Ferrara, Mushtaq Ahmad, Mehrnoosh Ostovar, Johnathan V. Matlock, and Varinder K. Aggarwal*

In memory of Gilbert Stork

Abstract: A convergent, nine-step (LLS), enantioselective synthesis of α -cyclopiazonic acid and related natural products is reported. The route features a) an enantioselective aziridination of an imine with a chiral sulfur ylide; b) a bioinspired (3+2)-cycloaddition of the aziridine onto an alkene; and c) installation of the acetyltetramic acid by an unprecedented tandem carbonylative lactamization/N-O cleavage of a bromoisoxazole.

Indole alkaloids have long been a source of inspiration for the development of new synthetic methods and strategies. α-Cyclopiazonic acid (α -CPA, **1**) is a prenylated indole alkaloid produced by a number of Penicillium species including P. commune, P. griseofulvum, and P. camemberti.[1] It is a potent inhibitor of Ca²⁺-dependent ATPase (SERCA) which prevents calcium reuptake in muscle.[2] In addition to its significant biological activity, α-CPA-producing fungi are found in cheese, meat, and other dietary products, making it important to the food industry.

Several structurally related natural products have been identified (Figure 1): iso- α -cyclopiazonic acid (2),^[3] α -CPA imine (3),[1b] speradines A-D,[4] and aspergillines A-E,[5] all sharing a 3-acetyltetramic acid unit.

Biosynthetically, α-CPA is derived from L-tryptophan (Figure 1B). [1a,b,6] The tetramic acid is assembled at an early stage followed by several alkylations to give β-cyclopiazonic acid (β -CPA, 4), a direct biosynthetic precursor of α -CPA. Flavin-mediated oxidation of β-CPA and subsequent cyclization give α -CPA.^[6f]

Four total syntheses of α -CPA have been published (Figure 2 A).^[7] They all share the same end-game strategy,

Supporting information and the ORCID identification number(s) for the author(s) of this article can be found under https://doi.org/10. 1002/anie.201712065.

© 2018 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly

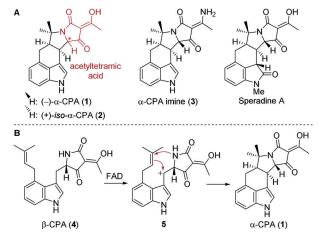


Figure 1. A) α -CPA and related natural products. B) Biosynthesis of α -

in which the tetramic acid unit is installed by a Dieckmann condensation, forming the C6-C7 bond. Kozikowski^[7a] and Natsume^[7b] constructed the C-D rings in a stepwise manner, but with low diastereoselectivity. Knight developed an elegant cationic cascade, in which acyclic precursor 9 was converted into indole 6 with high stereocontrol, [7c,d] although Scherkenbeck found that the same substrate cyclized to give a 1:1 mixture of diastereomers across the CD ring junction under slightly different conditions.^[7e,f]

In our retrosynthetic approach to α -CPA we considered a different, bioinspired strategy (Figure 2B). We were attracted by the possibility of using an aziridine 13 as a precursor to the zwitterionic intermediate 12 that would participate in a (3+2)-cycloaddition to construct the C-D ring system. Whilst this type of (3+2)-cycloaddition has been reported for the construction of pyrrolidines, [8] its application in total synthesis is much rarer. [9] Aziridine 13 could be assembled from simple building blocks 14 and 15 using our asymmetric sulfur ylide methodology.^[10] We envisaged using an isoxazole as a masked 1,3-dicarbonyl group^[11] attached to the sulfur ylide. A further attractive feature of this approach is that the ylide could carry all the carbons and functionality required for making rings C and D. We would then have to build ring E by N-C8 bond formation, rather than the C6-C7 bond, which is commonly used to construct tetramic acids.

We began our synthesis by targeting the imine building block 14 which was obtained in 4 steps from commercially

^[*] Dr. O. Zhurakovskyi, Dr. Y. E. Türkmen, L. E. Löffler, V. A. Moorthie, C. C. Chen, M. A. Shaw, M. R. Crimmin, M. Ferrara, Dr. M. Ahmad, Dr. M. Ostovar, J. V. Matlock, Prof. Dr. V. K. Aggarwal School of Chemistry, University of Bristol Cantock's Close, Bristol, BS8 1TS (UK) E-mail: v.aggarwal@bristol.ac.uk

Figure 2. A) Previous syntheses of α -CPA. B) Our retrosynthetic analy-

available indole 16 (Scheme 1 A). Suzuki cross-coupling of aryl bromide 16 with allyl boronic ester followed by Ntosylation gave indole 17. Cross-metathesis of the terminal alkene 17 in neat 2-methyl-2-butene [12] delivered 18 in good yield. Initial attempts to affect a one-step prenylation of 16 under various conditions led to substantial prenylboration of the aldehyde giving alcohol 19. The aldehyde 18 was converted into the N-nosyl^[13] imine 14, thus completing the synthesis of the indole fragment.

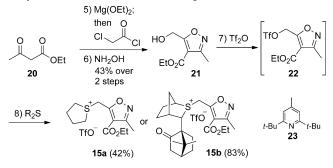
Sulfonium salts 15 a,b were prepared from known alcohol 21^[14] by a two-step sequence via triflate 22 (Scheme 1B). The use of the triflate instead of a corresponding bromide resulted in 1) much faster alkylations, and 2) the sulfonium salts precipitating directly from the ethereal solvent, permitting straightforward isolation by simple filtration.^[15]

Our initial synthetic campaign was performed with an achiral sulfonium salt 15 a to evaluate the viability of the route (Scheme 2). Reaction of imine 14 with an ylide derived from 15a proceeded smoothly and delivered aziridine 24 in good yield (72%) and diastereoselectivity (trans/cis 9:1). Trans-24 was prone to rapid isomerization into cis-24 in CDCl3 or on silica, and so was used crude. Notably, compound 24 already contains all the carbon atoms present in α -CPA.

We then explored the bioinspired cycloaddition of 24 and tested a range of Lewis and Brønsted acids (see the Supporting Information, SI), and found that treatment of CH₂Cl₂ solutions of 24 with 2 equiv of In(OTf)₃ or 0.1–1 equiv

A. Synthesis of the Indole Building Block 19

B. Synthesis of the Isoxazole Ester Building Blocks 15a-b



Scheme 1. Reagents and conditions: 1) Allyl-Bpin, Pd(dppf)Cl₂, KOH, THF-H₂O, 65 °C; 2) NaH, TsCl, DMF, 0 to 23 °C, 91 % over 2 steps; 3) 2-methyl-2-butene, Grubbs 2nd gen cat., 23 °C, 92 %; 4) NsNH₂, Ti(OEt)₄, CH₂Cl₂, 23 °C, 80%; 5) Mg(OEt)₂, PhH-EtOH, 23 °C, then 2chloroacetyl chloride, MeCN-PhH-EtOH, 0 to 23 °C, 44%; 6) NH₂OH·HCl, NaOAc, EtOH, reflux, 98%; 7) Tf₂O, 23, CH₂Cl₂, 0°C; 8) R_2S , Et_2O , 0°C, 42% for 15a, 83% for 15b. Pin=pinacolato, dppf=1,1'-bis(diphenylphosphino) ferrocene, Ts=4-toluenesulfonyl, Ns = 4-nitrobenzenesulfonyl, Tf = trifluoromethanesulfonyl.

of TfOH triggered the desired reaction. This gave pyrrolidine 25 as a mixture of diastereomers at C-11 (d.r. 3:1, in favor of the desired cis-isomer), from which the desired cis product was isolated as a single isomer in 24% yield by crystallization from MeCN-H₂O. The nosyl group was removed with PhSNa and the ester was hydrolyzed with LiOH yielding amino acid 26. Subjecting 26 to a standard amide coupling conditions (HATU, DIPEA, DMF) resulted in formation of lactam 27. Subsequent hydrogenolysis of the N-O bond under Pd catalysis gave N-Ts α-CPA imine 28 in 80% yield, which was then hydrolyzed^[7b] to give (\pm) - α -CPA (1) in 60% yield (dr 3.8:1). The racemic synthesis of 1 was thus achieved in 11 steps (longest linear sequence).

Unexpectedly, the attempted enantioselective campaign met with failure. The use of the chiral sulfonium salt 15b gave the desired aziridine 24 but with poor diastereo- and enantioselectivity (dr 1:0.9, er 40:60). We believe that the ylide derived from sulfonium salt 15b behaves as a stabilized rather than a semi-stabilized ylide and so reacts reversibly with the imine 14, resulting in low stereocontrol. [10a,c] We therefore considered alternative isoxazole substrates 31a-c (Scheme 3) bearing a less anion-stabilizing group (bromide in



5213773, 2018, 5, Downloaded from https://onlinelibtrary.wiely.com/doi/10.1002/anie.201712055 by Test, Wiley Online Library on [2010/2023]. See the Terms and Conditions (https://onlinelibtrary.wiely.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons License

Scheme 2. Reagents and conditions: 1) Cs_2CO_3 , CH_2Cl_2 , -40°C, 72%; 2) $In(OTf)_3$, CH_2Cl_2 , -78 to 23°C, 24%; 3) PhSNa, DMF, 23°C, 73%; 4) LiOH, THF-MeOH-H₂O, 23°C, 83%; 5) HATU, DIPEA, DMF, 23°C, 81%; 6) H₂ (1 atm), $Pd(OH)_2/C$, MeOH, 80%; 7) KOH, EtOH, 65°C. HATU = N-[(dimethylamino)-1H-1,2,3-triazolo-[4,5-b]pyridin-1-ylmethylene]-N-methylmethanaminium hexafluorophosphate N-oxide, DIPEA = N,N-diisopropylethylamine.

Synthesis of Bromoisoxazole Building Blocks 31a-c

Scheme 3. Reagents and conditions 1) acetaldoxime, NaClO, $CH_2Cl_2H_2O$, 0 to 23 °C, 65 %; 2) NBS, H_2SO_4 , AcOH, 110 °C, then NaOH workup, 85 %; 3) Tf_2O , **23**, CH_2Cl_2 , 0 °C; 4) R_2S , Et_2O , 0 °C, 93 % for **31 a**, 77 % for **31 b**, 47 % for **31 c**. NBS = N-bromosuccinimide.

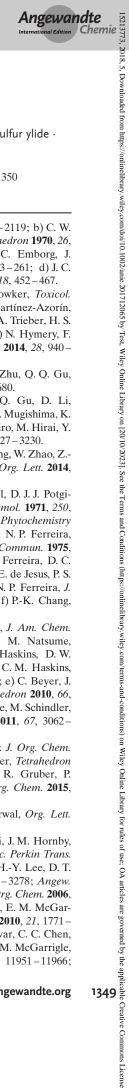
place of the ester). The bromine atom could also conveniently serve as a handle for a Pd-mediated carbonylative coupling.

Our second-generation synthesis of α -CPA began with the synthesis of bromoisoxazole sulfonium salts $\bf 31\,a-c$ as shown in Scheme 3. Alcohol $\bf 29$ was prepared in 2 steps from propargyl alcohol using a modified literature procedure. Triflation of $\bf 29$ followed by the nucleophilic substitution with a range of sulfides delivered the desired salts $\bf 31\,a-c$ in moderate to excellent yields.

Aziridination of imine **14** with ylides derived from **31a–c** afforded aziridine **32** in good yields, and as before, under exceptionally mild conditions (Scheme 4). The camphorderived salt **31b** performed better than the isothiocineolederived salt **31c** giving the aziridine **32** with good diastereoselectivity (*trans/cis* 9:1) and excellent enantioselectivity (er 98:2 for *trans*, 89:11 for *cis*). The high enantioselectivity

Second Generation (Enantioselective) Synthesis

Scheme 4. Reagents and conditions: 1) K₂CO₃, MeCN, -20°C, 56% (*trans/cis* 9:1, er 98:2 [*trans*], 89:11 [*cis*]); 2) TfOH, CH₂Cl₂, -55 to 10°C, 50% (dr 3.5:1, er 98:2); 3) PhSH, K₂CO₃, 18-crown-6, MeCN, 23°C, 63%; 4) CO (1 atm), Pd(OAc)₂, *n*-BuPAd₂, DABCO, DMSO, 120°C, 80%; 5) Cs₂CO₃, MeOH-THF-H₂O, 65°C, 70%. Ad = adamantyl, DABCO = 1,4-diazabicyclo[2.2.2]octane.



provided validation of our hypothesis: the ylide with the less electron-withdrawing bromine atom is now behaving as a semi-stabilized ylide, rendering betaine formation the enantiodetermining step.

As with aziridine 24, trans-aziridine 32 was prone to isomerization into cis-32, and thus was used without purification. Treatment of crude 32 with TfOH gave pyrrolidine 33 as a 3.5:1 mixture of diastereomers^[17] at C-11 in favor of the desired cis-isomer, in 50% yield with complete enantiospecificity (er 98:2).[18] Deprotection of the diastereomeric mixture with PhSH/K₂CO₃ gave amine 34, at which point the diastereomers were separated. We were initially concerned about the next Pd-catalyzed carbonylation-amide formation due to the severe angle strain inherent in the fused bicyclic isoxazole 27.[19] However, we were delighted to find that treatment of 34 with Pd(OAc)2 under an atmosphere of CO in the presence of DABCO and n-BuPAd₂^[20] triggered a reaction cascade leading directly to the formation of N-Ts α -CPA imine 28 in 80% yield. The cascade involves palladiumcatalyzed carbonylation, acylation, followed by reduction of the N-O bond in situ, [21] facilitated by the inherent angle strain of the fused unsaturated ring system 27. Presumably, the facility of the cyclization stems from ready formation of the undistorted amino-acyl palladium intermediate, before ring strain is introduced through the subsequent reductive elimination. Hydrolysis of N-Ts species 28 under basic conditions^[7e] in MeOH-THF-H₂O (10:10:1) provided a mixture of (-)- α -CPA (1) and (+)-iso- α -CPA (2) (dr 2.5:1) which was separated by reverse-phase prep-HPLC.

Synthetic (-)- α -CPA was identical in all respects to the natural material, including TLC, LCMS, HRMS, NMR and optical rotation^[3] data (see SI). When the reaction was performed under strictly anhydrous conditions, α-CPA imine (3) was the major product. This constitutes the first direct synthesis of α -CPA imine: the previous method relied on the amination of α-CPA itself.^[7e] This completed our synthesis of the α -CPA family.

In summary, we have achieved an enantioselective total synthesis of (–)- α -CPA and (+)-iso- α -CPA in 9 steps (LLS) from commercially available materials (13 total steps). The route is convergent with the key asymmetric aziridination bringing together the two halves of the molecule with high stereoselectivity and with all the functionality required to complete the target. Additional features of the sequence include 1) a bio-inspired intramolecular alkene-aziridine (3+2)-cycloaddition to assemble a polysubstituted pyrrolidine; and 2) a one-pot carbonylative lactamization/isoxazole cleavage to give an acetyltetramic acid. The latter represents a novel route to tetramic acids which could have broader applications in synthesis.

Acknowledgements

We thank EPSRC (EP/I038071/1), H2020 ERC (670668) and the University of Bristol for financial support. We also thank Siying Zhong for DFT calculations and Dr. Hazel Sparkes for X-ray analyses.

Conflict of interest

The authors declare no conflict of interest.

Keywords: (3+2)-cycloaddition · aziridination · sulfur ylide · total synthesis \cdot α -cyclopiazonic acid

How to cite: Angew. Chem. Int. Ed. 2018, 57, 1346-1350 Angew. Chem. 2018, 130, 1360-1364

- [1] a) C. W. Holzapfel, Tetrahedron 1968, 24, 2101-2119; b) C. W. Holzapfel, R. D. Hutchison, D. C. Wilkins, Tetrahedron 1970, 26, 5239-5245; c) K. Hermansen, J. C. Frisvad, C. Emborg, J. Hansen, FEMS Microbiol. Lett. 1984, 21, 253-261; d) J. C. Frisvad, Arch. Environ. Contam. Toxicol. 1989, 18, 452-467.
- [2] a) R. T. Riley, D. E. Goeger, H. Yoo, J. L. Showker, Toxicol. Appl. Pharmacol. 1992, 114, 261-267; b) F. Martínez-Azorín, FEBS Lett. 2004, 576, 73-76; c) K. Moncoq, C. A. Trieber, H. S. Young, J. Biol. Chem. 2007, 282, 9748-9757; d) N. Hymery, F. Masson, G. Barbier, E. Coton, Toxicol. In Vitro 2014, 28, 940-
- [3] A. Q. Lin, L. Du, Y. C. Fang, F. Z. Wang, T. J. Zhu, Q. Q. Gu, W. M. Zhu, Chem. Nat. Compd. 2009, 45, 677-680.
- [4] a) X. Ma, J. Peng, G. Wu, T. Zhu, G. Li, Q. Gu, D. Li, Tetrahedron 2015, 71, 3522 – 3527; b) M. Tsuda, T. Mugishima, K. Komatsu, T. Sone, M. Tanaka, Y. Mikami, M. Shiro, M. Hirai, Y. Ohizumi, J. Kobayashi, *Tetrahedron* **2003**, *59*, 3227 – 3230.
- [5] M. Zhou, M.-M. Miao, G. Du, X.-N. Li, S.-Z. Shang, W. Zhao, Z.-H. Liu, G.-Y. Yang, C.-T. Che, Q.-F. Hu, et al., Org. Lett. 2014, 16, 5016-5019.
- [6] a) J. C. Schabort, D. C. Wilkins, C. W. Holzapfel, D. J. J. Potgieter, A. W. Neitz, Biochim. Biophys. Acta Enzymol. 1971, 250, 311-328; b) C. W. Holzapfel, D. C. Wilkins, *Phytochemistry* 1971, 10, 351-358; c) P. S. Steyn, R. Vleggaar, N. P. Ferreira, G. W. Kirby, M. J. Varley, J. Chem. Soc. Chem. Commun. 1975, 465-466; d) R. M. McGrath, P. S. Steyn, N. P. Ferreira, D. C. Neethling, *Bioorg. Chem.* **1976**, 5, 11-23; e) A. E. de Jesus, P. S. Steyn, R. Vleggaar, G. W. Kirby, M. J. Varley, N. P. Ferreira, J. Chem. Soc. Perkin Trans. 1 1981, 3292-3294; f) P.-K. Chang, K. C. Ehrlich, I. Fujii, Toxins 2009, 1, 74-99.
- [7] a) A. P. Kozikowski, M. N. Greco, J. P. Springer, J. Am. Chem. Soc. 1984, 106, 6873-6874; b) H. Murakate, M. Natsume, Heterocycles 1985, 23, 1111-1117; c) C. M. Haskins, D. W. Knight, Chem. Commun. 2005, 3162-3164; d) C. M. Haskins, D. W. Knight, Tetrahedron 2011, 67, 8515-8528; e) C. Beyer, J. Scherkenbeck, F. Sondermann, A. Figge, Tetrahedron 2010, 66, 7119-7123; f) W. R. C. Beyer, K. Woithe, B. Luke, M. Schindler, H. Antonicek, J. Scherkenbeck, Tetrahedron 2011, 67, 3062-3070.
- [8] a) A. L. Cardoso, T. M. V. D. Pinho e Melo, Eur. J. Org. Chem. 2012, 6479-6501; b) S. H. Krake, S. C. Bergmeier, Tetrahedron 2010, 66, 7337-7360; c) E. Martinand-Lurin, R. Gruber, P. Retailleau, P. Fleurat-Lessard, P. Dauban, J. Org. Chem. 2015, 80. 1414 - 1426.
- [9] G. Arena, C. C. Chen, D. Leonori, V. K. Aggarwal, Org. Lett. **2013**, 15, 4250 – 4253.
- [10] a) V. K. Aggarwal, J. P. H. Charmant, C. Ciampi, J. M. Hornby, C. J. O'Brien, G. Hynd, R. Parsons, J. Chem. Soc. Perkin Trans. 1 2001, 3159-3166; b) V. K. Aggarwal, I. Bae, H.-Y. Lee, D. T. Williams, Angew. Chem. Int. Ed. 2003, 42, 3274-3278; Angew. Chem. 2003, 115, 3396-3400; c) R. Robiette, J. Org. Chem. 2006, 71, 2726-2734; d) M. Arshad, M. A. Fernandez, E. M. McGarrigle, V. K. Aggarwal, Tetrahedron: Asymmetry 2010, 21, 1771 -1776; e) O. Illa, M. Namutebi, C. Saha, M. Ostovar, C. C. Chen, M. F. Haddow, S. Nocquet-Thibault, M. Lusi, E. M. McGarrigle, V. K. Aggarwal, J. Am. Chem. Soc. 2013, 135, 11951-11966;



5213773, 2018, 5, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.201712065 by Test, Wiley Online Library on [2010/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons Licensee

- f) Review: V. K. Aggarwal, M. D. Badine, V. A. Moorthie in Aziridines and Epoxides in Organic Synthesis (Ed.: A. K. Yudin), Wiley-VCH, Weinheim, 2006, pp. 1-34.
- [11] a) G. Stork, A. A. Hagedorn III, J. Am. Chem. Soc. 1978, 100, 3609-3611; b) P. G. Baraldi, A. Barco, S. Benetti, G. P. Pollini, D. Simoni, Synthesis 1987, 857 – 869.
- [12] A. K. Chatterjee, D. P. Sanders, R. H. Grubbs, Org. Lett. 2002, 4, 1939-1942.
- [13] a) T. Fukuyama, C. K. Jow, M. Cheung, Tetrahedron Lett. 1995, 36, 6373-6374; b) T. Kan, T. Fukuyama, Chem. Commun. 2004, 353 - 359
- [14] a) S. Gelin, M. Chabannet, Synthesis 1978, 448-450; b) C. Deshayes, M. Chabannet, S. Gelin, Synthesis 1984, 868-870; c) V. A. Moorthie, E. M. Mcgarrigle, R. Stenson, V. K. Aggarwal, Arkivoc 2007, 139-151.
- [15] E. Vedejs, D. A. Engler, M. J. Mullins, J. Org. Chem. 1977, 42, 3109 - 3113.
- [16] a) S. Chimichi, M. Boccalini, B. Cosimelli, F. Dall'Acqua, G. Viola, Tetrahedron 2003, 59, 5215-5223; b) S. Al-Busafi, M. Al-Belushi, K. Al-Muqbali, Synth. Commun. 2010, 40, 1088-1092; c) E. Aktoudianakis, G. Chin, B. K. Corkey, J. Du, K. Elbel, R. H. Jiang, T. Kobayashi, R. Lee, R. Martinez, S. E. Metobo, et al., US 2014/0336190 A1, 2014.
- [17] Although the diastereoselectivity might appear moderate, it should be noted that the use of an ester group (32, $R = CO_2Et$) in place of the isoxazole resulted a 10:1 product ratio in favor of the undesired trans-isomer highlighting the sensitivity of the diastereomeric ratio to the substituent.
- [18] Cis- and trans-aziridines 32 converted into pyrrolidine 33 with the same diastereoselectivity and enantiospecificity but reacted

- at different rates. The reaction of trans-32 was cleaner, faster and higher yielding (50-60%) than that of cis-32 (30%).
- [19] Molecular modelling (B3LYP 6-311G(d)) shows that the ψ_1/ψ_2 angles in 27 should be close to 146° and 135°, respectively, which deviates substantially from the 124° and 117° in the strain-free system:

- [20] S. Guo, L. Tao, F. Wang, X. Fan, Chem. Asian J. 2016, 11, 3090-
- [21] The source of the two hydrogen atoms is intriguing. Exclusion experiments showed that either CO or DABCO could independently reduce the N-O bond of 27 under the reaction conditions (see SI, Table 7) but the reduction with CO was much cleaner. It is possible that adventitious H2O participates in a water-gas shift with CO generating CO2 and H2 which then reduces the N-O bond.

Manuscript received: November 23, 2017 Accepted manuscript online: December 19, 2017 Version of record online: January 9, 2018