# Experimental investigation on environmental control of a 50-person mine refuge chamber

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# 14 Abstract

15 Air quality and thermal environment of mine refuge chamber (MRC) are very important to determine the 16 physical safety of refugees. Accurately assessing the environmental load and taking reasonable measures are critical 17 to achieve the environmental control goals of MRC. In order to evaluate the metabolic parameters of occupants and the effectiveness of environmental control measures in a MRC, in this research, 50 adult men entered a MRC 18 19 laboratory for an 8-hour test. During the test, the compressed O2 cylinders and air purification devices were used to 20 ensure the indoor air quality. The possibility of using chemical adsorbents to passively scrub CO<sub>2</sub> and the performance 21 of dehumidification by mine compressed air (MCA) were also investigated by simulation experiments. The results 22 indicated that: (1) The per capita metabolic rates of  $O_2$ ,  $CO_2$  and heat during the refuge process are 0.34–0.37 L/min, 23 0.34 L/min and 117~128 W, respectively. (2) When Ca(OH)<sub>2</sub> particles are used as CO<sub>2</sub> adsorbent, the air purification 24 device has both dehumidification and  $CO_2$  scrubbing functions, and three air purification devices could make the 25 CO<sub>2</sub> concentration below 0.8% with the relative humidity below 76%. When Ca(OH)<sub>2</sub> particles are packaged to 26 passively scrub CO<sub>2</sub>, the amount of adsorbent may increase significantly. (3) When MCA is used for dehumidification 27 in a MRC, the air volume of  $0.15 \text{ m}^3/\text{min}$  per capita could maintain the relative humidity close to 60%. (4) In the 28 early stage of disaster avoidance, the indoor ambient temperature rises rapidly within 1 h followed by a slight increase.

29 Keywords: Mine refuge chamber; human metabolic rate; air quality; ambient temperature; relative humidity

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# 31 **1 Introduction**

32 With the increase of energy resource consumption and demand, the depth of commercial exploitation of mines 33 worldwide has also been increasing (Dou et al., 2020). In China, by the end of 2020, there are about 4,700 coal mines and 32,000 non-coal mines. Potential accidents such as explosion, fire, collapse can make underground mining to be 34 35 one of the most dangerous production activities (Onifade, 2021). Mine refuge chamber (MRC) is constructed in the 36 underground surrounding rock along the roadway and relatively isolated from the roadway environment, which provides breathable air, water and supplies in the event of an emergency within 96 h when miners are unable to escape 37 38 (Halim and Brune, 2019). MRC system has been adopted in underground mines in many countries, such as the United 39 States (Katherine et al., 2011), the United Kingdom (Brenkley et al., 2013), Chile (Mejias et al., 2014), Indonesia (Paul et al., 2019), etc.. Some applications can be seen in Fig. 1. 40





(a) A 30-person MRC in UK (Brenkley et al., 2013)

(b) A 50-person MRC in USA (Trackemas et al., 2015)





(c) A 80-person MRC in China (Li *et al.*, 2013) (d) A 300-person MRC in Indonesia (Paul et al., 2019) Fig. 1. Interior scenes of some MRCs in different countries.

Nomenclature
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A	Body surface area, m <sup>2</sup>	Subscrip	ts
с	Volume concentration	0	Oxygen
G	The O <sub>2</sub> supply rate, m <sup>3</sup> /min	с	Carbon dioxide
h	Height of people, m	Acronyn	15
M	Heat metabolic rate, W/m <sup>2</sup>	AT	Ambient temperature
п	Number of people in a MRC	APD	air purification device
q	Per capita heat metabolic rate, W	ISRT	initial surrounding rock temperature
V	Volume of a MRC, m <sup>3</sup>	MCA	Mine compressed air
v	Per capita metabolic rate	MRC	Mine refuge chamber
w	Weight of people, kg	PCM	Phase change materials
Gree	ek symbols	RH	Relative humidity
τ	Unit time, min		

During the refuge process in a MRC, it is recognized that the physiological metabolism of personnel will 6 7 consume O2 accompanied by the production of heat, water vapor and harmful gases such as CO2 and CO. This may 8 expose the MRC to a dangerous environment of hypoxia, hot, humid and high concentration harmful gases (Bauer, 9 et al., 2009, Yantek et al., 2019). Thus, in order to ensure environmental safety, some environmental control measures 10 should be considered including oxygen supply, harmful gas scrubbing, dehumidification and cooling. However, the underground power is normally cut off due to disasters and mine electrical devices must be explosion-proof. As a 11 12 consequence, it will become difficult to apply traditional air conditioning technology and equipment in MRCs.

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Ambient temperature (AT) in a MRC is affected by the initial surrounding rock temperature (ISRT) as well as 1 2 the heat transfer process between the surrounding rock and air (Yan et al., 2020a). When the ISRT is over 20 °C, the 3 cooling measure should be considered (Zhang et al., 2018). Mechanical ventilation is a conventional method for 4 controlling indoor relative humidity (RH) and AT, as well as air quality (Nitter et al., 2020). For the environmental 5 protection in a MRC, mine compressed air (MCA) generated from air compressors on surface and entering the MRC 6 through a protected MCA pipeline or a borehole is the most effective and economical method. However, it is not 7 practical to employ boreholes to deliver air to MRCs for some mines due to drilling costs, difficult terrain and surface 8 rights issues (Trackemas et al., 2015). In addition, the borehole is not suitable for MRCs with a buried depth of more 9 than 400 m. This is due to the reason that the earth-air heat exchange in the long vertical borehole can significantly 10 reduce the indoor AT while it only has a lag effect on the air humidity ratio, causing the indoor thermal discomfort 11 with a high RH (Gao et al., 2020). Several studies indicated that, when MCA is used in a MRC, the per capita air 12 volume of 42 L/min can meet the oxygen-supply demand (Shao et al., 2016), 100 L/min can meet the CO<sub>2</sub>-scrubbing 13 demand (Zhang et al., 2019) and the cooling demand in a MRC with an ISRT of 24 °C, 300 L/min can meet the 14 temperature control demand of a MRC with an ISRT of 27 °C within 96 h (Zhang et al., 2020).

15 For most underground mines in China, MCA is mainly sent into MRCs via a protected MCA pipeline. 16 Considering that the MCA pipeline may be damaged in the event of an accident, and it will take time to repair it. 17 Therefore, in addition to the MCA, at least one backup measure is needed for O<sub>2</sub> supply and harmful gas scrubbing. 18 In terms of  $O_2$  supply, the high-pressure oxygen cylinder is a more convenient and safe alternative measure compared 19 with chemical oxygen such as an oxygen candle. In terms of harmful gas scrubbing, methods of active chemisorption 20 through air purification device (APD) and passive chemisorption through chemical adsorbent curtains have been 21 proposed (Bauer et al., 2009), but previous studies mainly focus on the development of chemical adsorbent materials 22 and APDs. Some existing or newly synthesized chemical materials, including KO<sub>2</sub> oxygen plate (Gao et al., 2015), 23 g-C<sub>3</sub>N<sub>4</sub> foam/Cu<sub>2</sub>O ODs (Sun et al., 2019), soda lime (Yantek et al., 2019), modified soda lime (Du et al., 2018, Gai 24 et al., 2019), Ca(OH)<sub>2</sub> (Moreno et al., 2021) and amine-based modified porous material (Zhang et al., 2021), etc., have been used as  $CO_2$  adsorbents. Noble metal catalysts (e.g. PD-1, AU-1) and non-noble metal catalysts such as 25 Hopcalite have been selected to scrub CO (Jia et al., 2014). Taking into account the failure of underground power 26 27 supply, some air-conditioning equipment with only harmful gases scrubbing function, driven by battery-operated fans 28 (Bauer et al., 2009; Du et al., 2017), man-operated fans (Bo, 2014) or pneumatic fans (Zhang et al., 2017), have been 29 developed for low-temperature MRCs. Meanwhile, some air-conditioning equipment combined with cooling, 30 dehumidification, and air purification has also been developed for high-temperature MRCs (Yang et al., 2013). Zhang 31 et al. (2017) conducted a simulation experiment in a 50-person MRC in which APDs filled with Ca(OH)<sub>2</sub> particles 32 were used to scrub  $CO_2$ . Their experimental results showed that under the condition of a per capita  $CO_2$  metabolic 33 rate of 0.5 L/min, two selected APDs can make the indoor average CO<sub>2</sub> concentration below 0.8%.

34 As far as the AT control is concerned, it has been confirmed that for MRCs with an ISRT of above 27 °C, it is 35 difficult to achieve the temperature control requirement within 96 h by MCA alone. In recent years, some cooling 36 technologies, such as phase change material (PCM) cooling (Wu et al., 2012), liquid CO<sub>2</sub> cooling (Yang et al., 2013), ice storage cooling (Jia et al., 2015, Du et al., 2017, Shu et al., 2017, Xu et al., 2017), PCM combined with pre-37 38 cooling of the envelope (Yuan et al., 2017, Gao et al., 2017, Gao et al., 2018) and liquid air cooling (Yan et al., 2020b) 39 have been developed and applied widely. These cooling technologies have a commonality, that is, store the cold 40 sources at ordinary times and release the cold sources at a lower cost when people take refuge in order to achieve the 41 indoor temperature control in a MRC. PCM cooling is suitable for MRCs with low ISRT (Wu et al., 2012). Liquid 42 CO<sub>2</sub> cooling technology has a poor cooling effect in an environment with the AT above 32 °C accompanied by a risk 43 of leakage. Ice storage cooling technology has a wide range application prospect, but the supporting refrigeration 44 compressor is easily broken in a place with high temperature and humidity. In addition, the supporting circulating

fan needs to rely on other power, which reduces the reliability of the system. The cryogenic liquid air system is suitable for high-temperature MRCs, but with high technical requirement of maintaining liquid air at -195 °C and high costs from \$75 K to \$110 K (Yan *et al.*, 2020b).

4 It appears from the previous investigations that most of the work mainly focused on the environmental protection 5 characteristics of MCA used in MRCs, the development of adsorption materials, APD, and cooling technology to 6 cope with the failure of the MCA. However, there are few real human experiments carried out to investigate the 7 human metabolic parameters and the application performance of the environmental control measures in MRCs with 8 large capacity during the refuge process. In the current work, an 8-hour human experiment in a 50-person MRC will 9 be carried out to better understand the metabolic parameters of personnel during the refuge process and the 10 effectiveness of environmental protection measures in the MRC. Through monitoring the variation of gas 11 concentration, AT and RH with time, the metabolic parameters of personnel and the operating characteristics of air 12 conditioning devices in the MRC can be obtained. In addition, the possibility of passively scrubbing CO<sub>2</sub> through 13 Ca(OH)<sub>2</sub> bags will be also investigated through simulation experiments in a MRC.

# 14 2 Materials and Methods

# 15 2.1 Experimental environment

16 The experiment was carried out in a 50-person MRC laboratory located in Beibei District, Chongqing City of 17 China. Although the MRC laboratory is built on the ground, it can simulate the real interior scene of an underground 18 MRC for 50 people in full scale. The space of the MRC laboratory is divided into three areas including two transition 19 rooms and one living room. A two-door structure that opens outward is used, among them, the first door adopts a 20 protective closed door that can resist shock waves and block toxic gases, while the second door uses a closed door 21 which can only block toxic gases. The effective pass size of these doors is 0.8 m×1.6 m. The groove around the door 22 wall has a depth of not less than 0.2 m, and the wall is poured by concrete with a strength not lower than C30 to 23 ensure sufficient airtightness. The pouring thickness of the explosion-proof wall and the enclosed wall are 0.8 m and 24 0.3 m, respectively. The ground, arched roof and both-side walls of the MRC are poured with concrete, their thickness 25 are 0.4 m, 0.5 m and 0.3 m, respectively. To facilitate the passage of personnel and equipment, the same entrance is 26 set on both sides of the MRC. The living room occupies the arched space between the two enclosed walls with a 27 cross-section of 4 m in width, 3.5 m in height, 14 m in circumference, 13.2 m<sup>2</sup> in area, 17 m in length and 224.4 m<sup>3</sup> 28 in volume. There are 27 triple seats placed near one side wall of the room. A step is constructed with a steel plate on 29 the wall near the seat above the ground 1.2 m, which is used to place 10 sets of MCA self-rescue devices for 50 30 people. Along the bottom of the step, 25 lamp bases are installed for heating lamps (200 W each lamp) to simulate 31 the heat dissipation of human metabolism. The scene of the MRC laboratory is shown in Fig. 2.





## 1 **2.2** *Subjects*

The miners who benefit from a MRC are normally adult men, who are regarded as subjects in the current research. So subjects with age ranging from 28 to 45, and 80% of them are 30~40 years old. The height of most subjects ranges from 1.68 m to 1.76 m. The subjects are all healthy and they did not participate in alcohol activities before each experiment. Moreover, smoking and drinking are forbidden during the experiment. During the test period, adequate mineral water, compressed biscuits, and beef jerky, as well as two portable toilets are prepared to facilitate the diet and excretion of subjects.

#### 8 **2.3** Instruments and equipment

9 Before the experiment, 8 oxygen cylinders with 40 L and 20 MPa are prepared in the transition room to supply 10  $O_2$  gas. The structure principle of the  $O_2$  supply system is shown in Fig. 3. The  $O_2$  gas flow rate is adjusted by a 11 control cabinet and measured by a float flowmeter. There are three devices combined with functions of air purification, 12 dehumidification and cooling to protect the environment in the living room. The working principle of these devices 13 is to divert polluted air from the living room into the device through a circulating fan and sequentially pass through 14 an air purification module using chemical adsorption and a cooling and dehumidifying module using evaporative 15 cooling. Then the treated fresh air returns to the living room from the air outlet of the circulating fan, the structure principle of the APD is illustrated in Fig. 4. The air inlet is a rectangular surface with 0.4 m $\times$ 0.6 m, and the air outlet 16 17 is a circular surface with a radius of 0.05 m. It is worth mentioning here that the device is equipped with two 18 circulating fans. One is powered by a  $CO_2$  airflow, the liquid  $CO_2$  from the compressed cylinders is decompressed 19 twice to achieve phase change refrigeration for a MRC that requires cooling measure. And the decompressed CO<sub>2</sub> 20 airflow enters the circulating fan to drive it works, then is discharged to the outdoors through the draft tube. The other 21 fan powered by MCA or compressed air cylinders is mainly used in a MRC where the cooling measure is not required.



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Fig. 3 Structure principle of the O<sub>2</sub> supply system Fig. 4. Structure principle of the APD.

24 The experiment was carried out on a certain day in January, from 9:30am to 6:30pm. During the experiment, the 25 outdoor AT was 10 °C ~ 15 °C. It has been preliminarily evaluated that the AT in the living room will not exceed 30 °C 26 without cooling requirement and the CO concentration will be below 24 ppm without scrubbing measure during the 27 experiment. Therefore, compressed liquid CO<sub>2</sub> were not used for refrigeration and CO adsorbents were not used to 28 scrub CO, only CO<sub>2</sub> adsorbents were added to the device to scrub CO<sub>2</sub>. The circulating fan of the device was driven 29 by a compressed airflow with 0.3 MPa from an air compressor. The air velocity at the outlet of the circulating fan 30 was  $10\pm0.5$  m/s measured by an anemometer. Considering the chemical adsorbent used to scrub CO<sub>2</sub> in common for 31 MRCs, the  $CO_2$  adsorbent used in the experiment is  $Ca(OH)_2$  particles, which are made into a curtain shape so that 32 they can be quickly filled or replaced. Each curtain contains 3 kg  $Ca(OH)_2$  particles, which must be stored in a 33 vacuum seal before use. Each time the adsorbent needs to be added or replaced, each device consumes five curtains.

# 1 2.4 Data collection

In the living room, three measuring points for CO<sub>2</sub>, O<sub>2</sub>, CO and CH<sub>4</sub>, as well as two measuring points for RH are arranged on both sides of the wall about 1.5 m above the ground, 5 measuring points for AT are arranged at 1 m to 2.5 m height above the ground, the distribution of subjects, devices, and measuring points is shown in Fig. 4, information of sensors used for different parameters can be seen in Table 1.





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#### 8 Table 1. Information of sensors used for different parameters

(a) Parameter test platform

Parameters	Instrument/Model	Range	Accuracy	Manufacturer
CO <sub>2</sub> concentration	Mine infrared carbon dioxide sensor, GRG5H	0~5%	0.1%	
O <sub>2</sub> concentration	Mine oxygen sensor, GYH25	0~25%	0.1%	CCTEG Chongqing
CH <sub>4</sub> concentration	Mine methane sensor, KG9701B	0~4.00%	0.1%	Research Institute
CO concentration	Mine carbon monoxide sensor, GTH1000	0~1000 ppm	1 ppm	Co., Ltd.
RH	Mine temperature & humidity sensor, GWSD50/100	$0 \sim 100\%$	0.1%	
AT	Platinum thermal resistance, PT100	-100 ~ 300 °C	0.1 °C	ELECALL Co., Ltd.

The data monitored by these sensors is uploaded to a data acquisition system platform through multi-channel

10 recorders. The scene of the platform and the interface of the data acquisition system are shown in Fig. 5 (a) and (b),

11 respectively.

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(b) Interface of the acquisition system

Fig. 5. Scene of the platform and the interface of the data acquisition system.

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# 2.5 Experimental procedure

During the experiment, the subjects are in a sitting or light work state. To determine the per capita metabolic rate when people take refuge, the indoor  $CO_2$  and CO concentrations are in a natural rising state at the early stage of the experiment. When the average  $CO_2$  concentration is close to 1%, the  $CO_2$  absorbent will be added to the APDs to scrub  $CO_2$  gas. During the experiment, the APDs went through three different operating conditions to test their performance, and the  $O_2$  supply rate was adjusted in time to adapt to changes in  $O_2$  concentration. Fig. 6 shows the experimental scene at a certain moment after the heater lamps are switched on.



Fig. 6. Experimental scene of 50-person refuge test.

10	The main operating steps during the experiment are as follows:
11	(1) Before the experiment, open the data acquisition system for 0.5 h to obtain initial environmental parameters.
12	(2) All subjects quickly enter the living room, then close airtight doors on both sides.
13	(3) Open the $O_2$ supply system and adjust the flow to 40 L/min.
14	(4) Turn on the 3 sets of devices without filling the CO <sub>2</sub> adsorbent to stir to make the air distribution more even.
15	(5) At about 1.2 h, adjust the O <sub>2</sub> supply flow to 30 L/min, since the indoor O <sub>2</sub> concentration rises faster.
16	(6) At about 2 h, add CO <sub>2</sub> adsorbent to the $1^{\#}$ and $3^{\#}$ devices, since the CO <sub>2</sub> concentration is close to 1%.
17	(8) At about 4.8 h, replace CO <sub>2</sub> adsorbent in the $1^{\#}$ and $3^{\#}$ devices, and add CO <sub>2</sub> adsorbent to the $2^{\#}$ device.
18	(7) At about 5.3 h, adjust the O <sub>2</sub> supply flow to 20 L/min, because the O <sub>2</sub> concentration keeps rising.
19	(9) At about 6.6 h, adjust the O <sub>2</sub> supply flow to 10 L/min, and turn on the heating system.
20	(10) When the test lasts about 8.5 h, end the experiment and save the recorded data.
21	3 Results

# 22 **3.1** Oxygen concentration

Fig. 7 plots the variation of O<sub>2</sub> concentration in the MRC with time under four different operating conditions, i.e., O<sub>2</sub> supply rates of 40 L/min within 0~1.2 h, 30 L/min within 1.2~5.3 h, 20 L/min within 5.3~6.6 h, and 10 L/min within 6.6~8.5 h, respectively. The corresponding per capita O<sub>2</sub> supply rates are 0.8, 0.6, 0.4 and 0.2 L/min. From Fig. 7, it can be observed that, under the four different O<sub>2</sub> supply states, the O<sub>2</sub> concentration varies approximately

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- 1 linearly with time, which demonstrates that during the refuge process, the O<sub>2</sub> consumption rate of human metabolic
- 2 is relatively stable. The values of  $O_2$  concentration at points 2 and 3 are very close, while at point 1 is larger, the
- 3 maximum difference at a certain moment is close to 0.7%. This phenomenon may be caused by the fact that point 1
- 4 is far away from the subjects while both points 2 and 3 are located above the crowded areas and closer to the oxygen-5 supply diffuse tube. When the per capita  $O_2$  supply rate reaches 0.8 L/min, the  $O_2$  concentration increases quickly
- 6 with time, which means that the indoor  $O_2$  supply rate greatly exceeds the  $O_2$  consumption rate by personnel. When
- 7 the per capita  $O_2$  supply rate reduces to 0.5 L/min per person within 1.2~5.3 h, the  $O_2$  concentration curve is still on
- 8 the rise, but the growth gradient has decreased. This indicates that the per capita O<sub>2</sub> metabolic rate in the MRC is less
- 9 than 0.5 L/min. It should be noted that during  $2.5 \sim 3.5$  h, the indoor O<sub>2</sub> concentration increase rate is relatively slow
- 10 due to the subjects having lunch. When the per capita O<sub>2</sub> supply rate reaches 0.4 L/min per person within 5.3~6.6 h,
- 11 the  $O_2$  concentration curve changes smoothly. Whereas when the  $O_2$  supply rate reduces to 0.2 L/min per person, the
- 12 variation of  $O_2$  concentration with time becomes a downward trend. It can be deduced that the per capita  $O_2$  metabolic
- 13 rate ranges from 0.2 L/min to 0.4 L/min.



Fig. 7. O<sub>2</sub> concentration varies with time.

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Fig. 8. CO<sub>2</sub> concentration varies with time.

- Fig. 8 presents the variation of CO<sub>2</sub> concentration in the MRC with time under three different conditions, such 1 2 as three APDs operate without CO<sub>2</sub> adsorbent within 0~2 h, two APDs with CO<sub>2</sub> adsorbent operate within 2~4.8 h 3 and three APDs with CO<sub>2</sub> adsorbents operate within 4.8~8.5 h. It can be seen clearly that, for the cases of three APDs 4 are operated without CO<sub>2</sub> adsorbent, the CO<sub>2</sub> concentration approximately linearly varies with time. The CO<sub>2</sub> 5 concentration increased to about 1% after the 50 objects entered the living room for 2 h. When the CO<sub>2</sub> adsorbent is 6 filled into two APDs, the CO<sub>2</sub> concentration is maintained at a relatively stable value ranging from 0.9% to 1.1%, 7 which implies that two APDs used in this test can meet the control demand of CO<sub>2</sub> concentration for a 50-person 8 MRC. When three APDs filled with CO<sub>2</sub> adsorbent are operating in the 50-person MRC, the indoor CO<sub>2</sub> concentration 9 gradually decreases with time, the  $CO_2$  concentration value stabilizes at about 0.7% after 2 h.
- 10 **3.3** Carbon monoxide concentration





Fig. 9. CO concentration varies with time.

Fig. 9 demonstrates the variation of CO concentration with time in the MRC without CO scrubbing measure. It can be found that the CO concentration increases monotonically with time, but after 8.5 h, the CO concentration only increased by about 4 ppm. It can be inferred that the CO concentration in the MRC may increase to about 45 ppm at 96 h, the CO concentration will exceed the allowable value of 24 ppm by the current standard.

17 **3.4** *Methane concentration* 





Fig. 10. CH4 concentration varies with time.

Fig. 10 plots the variation of CH<sub>4</sub> concentration with time in the MRC without CH<sub>4</sub> scrubbing measure. It can be found that the CH<sub>4</sub> concentration oscillates between 0 and 0.1%. This phenomenon occurs mainly because the accuracy of the methane sensor is 0.1%, and there is almost no CH<sub>4</sub> gas produced in the process of human metabolism. According to the result, the changes in air quality caused by the CH<sub>4</sub> gas generated from human metabolism could be ignored within 96 h for a MRC.

## 6 **3.5** *Ambient temperature*

During the experiment, the MRC went through three periods with different thermal intensities. That is, there is only human metabolic heat within 0~2h, the chemical reaction heat between CO<sub>2</sub> and Ca(OH)<sub>2</sub> is increased within 2~6.3 h, and an additional heat of 5 KW is increased within 6.3~8.5 h by the 25 heating lamps.



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12 Fig. 11 presents the variation of AT in the MRC with time. It should be noted that the AT quickly rises to 14 °C 13 from 10 °C during 0.5 h followed by a slight increase of AT. From 1 h to 2 h, the average AT rises to 15.14 °C from 14 14.74 °C, the gradient of AT rise is about 0.40 °C/h. It can be found that the AT curve fluctuates slightly within 2~3 15 h. This could be the reason that when the CO<sub>2</sub> adsorbent is added to the devices and food is distributed to the subjects, the movement of some subjects indoors could lead to some interference to the measured value of these measuring 16 17 points. From 2 h to 6.3 h, due to the increase in chemical reaction heat between water vapor, CO<sub>2</sub>, and Ca(OH)<sub>2</sub>, the gradient of AT shows a slight increases compared to that within 1~2 h. The average AT rises to 17.25 °C at 6 h from 18 19 15.6 °C at 3 h, the gradient of AT rise is about 0.55 °C/h. When the additional 5 KW heat intensity suddenly is 20 increased by heating lamps at 6.5 h, the AT rises significantly in less than 0.5 h, then the gradient of AT rise drops again, the average AT rises from 19.74 °C at 7 h to 20.97 °C at 8.5 h, the gradient of AT rise is about 0.82 °C/h. 21

22 In the current experiment, there is no external ventilation, therefore, the heat generated by subjects and devices 23 is mainly absorbed by the indoor air and walls. It can be deduced from Fig.11 that when the heat intensity suddenly 24 increases, the AT will rise rapidly in a short period that is less than 0.5 h, then the AT increase rate will gradually 25 decrease until it tends to be a relatively balanced state. The reason for this phenomenon is that when the indoor heat 26 intensity increases suddenly, the heat exchange rate between the air and the wall is relatively small due to the small 27 temperature difference, except the heat absorbed by the wall, the excess heat is absorbed by the air with a relatively 28 small specific heat capacity so that the AT rises rapidly. As the AT increases, the temperature difference between the 29 air and the wall increases, and the heat absorbed by the wall increases until it is almost equal to the indoor heat 30 generation. It can be inferred from this that, in a MRC, the heat exchange between the surrounding rock and the air 31 has an important influence on the control effect of the AT, the gradient of AT rise is positively correlated with the 32 indoor heat intensity.

#### 1 **3.6** *Relative humidity*



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Fig. 12. RH in the MRC varies with time.

Fig. 12 plots the variation of RH in the MRC with time. It can be found that in the first two hours, the RH increased relatively quickly, from 76.2% to 79.5%. When two APDs filled with CO<sub>2</sub> adsorbents operate from 2 h to 4.8 h, the gradient of RH decreased slightly since the water vapor reacted with Ca(OH)<sub>2</sub> particles and are absorbed. The RH slowly increases to approximately 81% in nearly 3 h. When the three devices are used from 4.8 h, the RH in the MRC starts to drop, and the RH reaches approximately 76% within 3 h, meeting the requirement of RH control in the MRC. This means that APDs can control RH and CO<sub>2</sub> concentration at the same time for MRCs.

#### 10 4 Discussion

#### 11 **4.1** Metabolic parameter of refugees

12 The main control parameters of human metabolism such as the metabolic rates of  $O_2$ ,  $CO_2$  and heat can be 13 calculated from the experimental data.

When the O<sub>2</sub> gas is supplied only through compressed oxygen cylinders in the MRC, and the APDs are running without chemical adsorbents. According to the law of conservation of matter, there is

 $V\partial \frac{c_{\rm o}(\tau)}{\Box \tau} = G(\tau) - nv_{\rm o}(\tau) \times 10^{-3}$ <sup>(1)</sup>

17 where *V* is the volume of the MRC, m<sup>3</sup>;  $\tau$  is the unit time, min;  $c_0$  is the O<sub>2</sub> concentration; *G* is the O<sub>2</sub> supply rate, 18 m<sup>3</sup>/min; *n* is the number of people in the MRC;  $v_0$  is the per capita O<sub>2</sub> metabolic rate, L/min.

19 Thus, the per capita  $O_2$  metabolic rate can be calculated as

$$v_{o}(\tau) = \left[ G(\tau) - V \partial \frac{c_{o}(\tau)}{\Box \tau} \right] \times 10^{3} / n$$
<sup>(2)</sup>

21 When there are no  $CO_2$  scrubbing measures in the MRC, the indoor  $CO_2$  concentration is only affected by the 22  $CO_2$  metabolism of personnel. According to the law of conservation of matter, there is

23  $V\partial \frac{c_{\rm c}(\tau)}{\Box \tau} = nv_{\rm c}(\tau) \times 10^{-3}$ (3)

24 where  $c_c$  is the CO<sub>2</sub> concentration;  $v_c$  is the per capita CO<sub>2</sub> metabolic rate, L/min.

25 Thus, the per capita CO<sub>2</sub> metabolic rate can be calculated as

26 
$$v_{\rm c}(\tau) = V \partial \frac{c_{\rm c}(\tau)}{n \Box \tau} \times 10^3$$
(4)

1 The heat metabolic rate is determined by the measured per capita  $O_2$  metabolic rate, the per capita  $CO_2$  metabolic 2 rate, and the body surface area, it can be indirectly calculated as follows (Zhai *et al.*, 2018)

3

5

6

9

23

 $M = (0.23v_{\rm c} + 0.77v_{\rm o}) \times 5.88 \times 60/A \tag{5}$ 

4 where *M* is the heat metabolic rate,  $W/m^2$ ; *A* is the body surface area,  $m^2$ .

The body surface area can be determined using the following formula (Shimazaki and Katsuta, 2019)

 $A = 0.2w^{0.425} \times h^{0.725} \tag{6}$ 

7 where w is the weight of people, kg; h is the height of people, m. For Chinese adult men, the average value is 8  $1.80 \sim 1.83 \text{ m}^2$ ;

## (1) O<sub>2</sub> metabolic rate

As far as the  $O_2$  metabolic rate is concerned, in the current experiment, the use of compressed oxygen bottles to supply oxygen is the only measure. Once the value of  $O_2$  supply at each stage is determined, then the average oxygen metabolic rate of each stage can be theoretically calculated. However, from the experimental results, the relative deviation of the three measurement points after 5.3 h is relatively large. Therefore, according to Eq. (2), it is more appropriate to use the experimental data of the first two stages before 5.3 h as the calculation basis. The oxygen supply flow rate is 40 L/min from 0 to 1.2 h and 30 L/min from 1.2 to 5.3 h, respectively. The indoor average  $O_2$ concentration of the three measuring points is 20.77% at 0 h, 21.53% at 1.2 h and 22.77% at 5.3 h, respectively.

17 From 0 to 1.2 h, the per capita  $O_2$  metabolic rate in the MRC can be calculated as

18 
$$v_{o}(\tau) = \left[ G(\tau) - V \partial \frac{c_{o}(\tau)}{\Box \tau} \right] / n = \frac{40 \times 1.2 \times 60 - 224.4 \times 10^{3} \times (21.53\% - 20.77\%)}{50 \times 1.2 \times 60} \approx 0.33 \text{ L/min}$$

19 From 1.2 to 5.3 h, the per capita O<sub>2</sub> metabolic rate in the MRC can be calculated as

20 
$$v_{o}(\tau) = \left[ G(\tau) - V \partial \frac{c_{o}(\tau)}{\Box \tau} \right] / n = \frac{30 \times (5.3 - 1.2) \times 60 - 224.4 \times 10^{3} \times (22.77\% - 21.53\%)}{50 \times (5.3 - 1.2) \times 60} \approx 0.37 \text{ L/min}$$

It can be found that the  $O_2$  metabolic rate increased slightly within 1.2~5.3 h compared with that within 0~1.2 h, this could be due to the diet of the subjects during this period.

#### (2) CO<sub>2</sub> metabolic rate

As far as the CO<sub>2</sub> metabolism rate is concerned, since the MRC has adopted air purification measures after 2 h, and several parameters of the air purification device such as purification efficiency are still unclear, the experimental results cannot be used as a valid reference value, according to Eq. (4). Therefore, values of CO<sub>2</sub> concentration without purification measures within  $0\sim2$  h are selected for per capita CO<sub>2</sub> metabolism. It can be found that the CO<sub>2</sub> concentration grows approximately linearly with time from 0 to 2 h, and the linear fitting formula is y=0.09+0.43x, with  $R^2=0.99$ . The indoor average CO<sub>2</sub> concentration of the three measuring points is 0.07% at 0h and 0.97% at 2 h. Thus, the per capita CO<sub>2</sub> metabolic rate in the MRC can be calculated as

31 
$$v_{\rm c}(\tau) = V \partial \frac{c_{\rm c}(\tau)}{n \Box \tau} = \frac{224.4 \times 10^3 \times (0.97\% - 0.07\%)}{50 \times 2 \times 60} \approx 0.34$$
 L/min

Thus, the per capita CO<sub>2</sub> metabolic rate is 0.34 L/min for Chinese people who are taking refuge in a MRC. Yang et al. (2020) observed that for adult men at lying and office activity levels, the CO<sub>2</sub> metabolic rates range from 0.23 to 0.34 L/min, which is more consistent with our experimental results.

#### 35 (3) Heat metabolic rate

36 The calculation method of the human metabolic heat in Eq. (5) takes into account the surface area of the human

1 body. But the surface area of each subject is different, so there is a slight gap in the amount of heat dissipation.

2 Therefore, when we calculate the average heat dissipation rate of subjects, we consider that all subjects have the same

metabolic parameter values and body surface area. Thus, when calculating the per capita heat metabolic rate in the
 MRC, according to Eq. (7), there is

5

 $q = M \cdot A = (0.23v_{\rm c} + 0.77v_{\rm o}) \times 5.88 \times 60$ 

(7)

6 Where q is the per capita heat metabolic rate, W.

Thus, from 0 to 1.2 h, the per capita heat metabolic rate in the MRC can be calculated as

7 8

From 1.2 to 5.3 h, the per capita heat metabolic rate in the MRC can be calculated as

10

14

9

*q*=(0.23×0.34+0.77×0.38) ×5.88×60≈128 W

q=(0.23×0.34+0.77×0.33) ×5.88×60≈117 W

11 Thus, for Chinese adult men, when they are taking in a MRC, the per capita heat metabolic rate ranges from 12 117 W to 123 W, which is consistent with the recommendation value of 113~134 W when people are resting in a 13 refuge alternative (Bernard *et al.*, 2018).

# (4) Sweat metabolic rate

Hirata *et al.* (2015) found that young men with an age of 20~30 exposed to the thermal environment with RH of 60% and temperature of 32.5 °C while standing for 3 h, the sweat metabolic rate is about 4.5 g/min. According to the result of Klein *et al.*, (2017), at an apparent temperature of 96.5 °F (35.83 °C), the moisture loss, from sweat and respiration, of 1.0 L/day for both the 78.5 kg and 111 kg humans with high body fat at an activity level of 1.0 met. During the refuge process, the activity levels of occupants in the MRC range from 0.8 to 1.0 met, the total moisture loss is less than 1 L/day per capita.

# 21 4.2 Air quality control in MRC

In terms of harmful gas scrubbing, the use of APDs must rely on external power, which will affect the reliability of the system to a certain extent. Although a method for passively scrubbing  $CO_2$  through lithium curtains has been proposed (Bauer *et al.*, 2009), it has not been verified. Therefore, it is worth discussing whether the direct placement of chemical adsorbents in the MRC to passive scrub  $CO_2$  meets the control requirement. A test was carried out in another MRC laboratory with a length of 20 m, a width of 4 m and a height of 3 m, as shown in Fig.13.



27 28

Fig. 13. remove CO<sub>2</sub> by adsorbent bags.

For more laboratory information, please refer to Zhang *et al.* (2020). During the test, the compressed CO<sub>2</sub> gas cylinders and diffuse pipes located on both sides of the room were used to simulate the CO<sub>2</sub> metabolism of 50 people,

- 1 and the per capita  $CO_2$  metabolic rate is 0.5 L/min. Before the test, the initial  $CO_2$  concentration of the MRC
- 2 laboratory is closed to 1.1%. Ca(OH)<sub>2</sub> particles used to scrub CO<sub>2</sub> are packed into a small sand cloth bag every 2 kg.
- 3 Respectively, 40, 80, and 130 bags are hung on four shelves, at times of 0 h, 0.2 h, and 0.6 h, to passively control
- 4  $CO_2$  concentration. There are five measuring points distributed in the room to monitor  $CO_2$  concentration in realtime.







Fig. 14. CO<sub>2</sub> concentration varies with time.

8 Fig. 14 plots the variation of CO<sub>2</sub> concentration with time. It can be observed that when 40 bags of CO<sub>2</sub> adsorbent 9 bags were used to scrub CO<sub>2</sub> gas in a 50-person MRC, the CO<sub>2</sub> concentration trends upward over time and the average 10 value rises to 1.65% from 1.1% within 0.3 h. When the CO<sub>2</sub> adsorbent bags increase to 80 bags at 0.3 h, the rising trend of the CO<sub>2</sub> concentration slows down, and the average CO<sub>2</sub> concentration rises to 1.82% from 1.65% within 11 12 0.35 h. When the CO<sub>2</sub> adsorbent bags increase to 130 bags at 0.65 h, the CO<sub>2</sub> concentration shows a downward trend 13 with time, and the average CO<sub>2</sub> concentration stabilizes at 1.52% after 0.4 h. According to the experimental result, it 14 can be imagined that if the CO<sub>2</sub> adsorbent bags continue to increase, the CO<sub>2</sub> concentration can be controlled below 1%. However, directly hanging the  $CO_2$  adsorbent bags will greatly increase the amount of the  $CO_2$  adsorbent, 15 16 compared to the use of APDs.

# 17 **4.3** Thermal environment control in MRC

## (1) RH control

19 From the experimental results, it can be seen that the RH control in a MRC can be achieved simultaneously with 20  $CO_2$  concentration when chemisorption is used to scrub  $CO_2$ . In contrast, for lower storage RH ( $\leq$ 75%), carbonation 21 contributes only by a small fraction to CO<sub>2</sub> capture, most of the CO<sub>2</sub> captured is probably dissolved or in the form of 22 bicarbonate ions in the physisorbed water present in Ca(OH)2 (Moreno et al., 2021). In addition, in an initial high-23 temperature MRC where a cooling measure is required, dehumidification can also be achieved simultaneously with 24 air cooling when refrigeration devices cool the circulating air through condensers (Yang et al., 2013). However, 25 compared with these dehumidification methods, MCA is also the simplest measure for dehumidification in a MRC. Its principle is that the air from the external environment enters a high-pressure storage tank after being compressed 26 27 by a compressor, and exchanges heat with the outside through the surfaces of the storage tank and conveying pipeline, 28 to realize the compression, cooling and dehumidification of the high-humidity air, then the fresh air with low RH 29 enters the MRC to dilute the high-humidity air to achieve the purpose of humidity control. The parameters of MCA 30 that meet the requirement of RH control in a MRC is worth studying.

31 A dehumidification test through MCA was carried out in the laboratory, as shown in Fig. 15. According to the

- result of Klein et al., (2017), the moisture loss of occupants in MRC is 0.86 L/day per capita for this test. An ultrasonic 1 2 humidifier with a water vapor dissipation capacity of 1.8 kg/h was used to simulate the moisture dissipation of 50 3 people, and a high-power fan was used to stir the indoor air to promote better dispersion. Compressed air provided 4 by an air compressor with a rated air volume of 10 m<sup>3</sup>/min was sent to the MRC laboratory from air outlets located 5 on both side walls. The volume rate of air supply is controlled by a pressure-reducing valve and measured by a vortex 6 flowmeter. A humidity sensor is placed in the center of the room above the ground 1m to obtain the real-time value 7 of RH. The AT and RH outside the MRC were 27°C-29°C and 74%-80%, respectively, during the period of this test. 8 The initial RH in the MRC ranges from 85% to 90%, the compressed air supply rates for the MRC were 250 m<sup>3</sup>/h,
- 9  $300 \text{ m}^3/\text{h}$ , and  $400 \text{ m}^3/\text{h}$ , respectively. The scene of the test is shown in Fig. 15.



Fig. 15. RH in MRC varies with time under different ventilation rates.



Fig. 16. RH in MRC varies with time under different ventilation rates.

14 Fig. 16 plots the variation of RH in the MRC with time under different ventilation rates. It can be found that 15 when the air supply rate is 250 m<sup>3</sup>/h, the RH drops from 88% to 85% within 1 h, and a relatively stable value is maintained. This basically satisfies the requirement of humidity control in the MRC, but the humidity should be 16 improved in terms of comfort. When the air ventilation rate is increased to 300 m<sup>3</sup>/h (0.1 m<sup>3</sup>/min per capita), the RH 17 in the MRC further drops and remains at about 68% from 89% in 1 h, the wet environment has been significantly 18 improved. When the air supply rate increases to 400 m<sup>2</sup>/h, the RH dropped from 88% to 61% in 1 h, which is close 19 20 to the relatively comfortable range. It should be noted that the RH does not decrease linearly with the air supply rate. 21 From the result of this test, it can be seen that in a MRC with compressors on the ground to provide MCA, when the per capita air volume is above 0.15 m<sup>3</sup>/min, it is easy to control the RH below 60%. 22

23 (2) AT control

1 When calculating the heat load in a MRC, although the heat metabolic rate of each individual is slightly different 2 under the same activity intensity, in terms of the heat metabolic rate during the refuge period, it is relatively reasonable 3 to take the value of 120~130 W per capita. In addition, when chemical absorbents are used to scrub CO<sub>2</sub>, an 4 exothermal chemical reaction heat of 40~50 W cannot be ignored (Halim et al., 2019). It is worth noting that the 5 dynamic heat transfer between the surrounding rock and the air in MRC has an important influence on temperature 6 control. Our previous studies showed that, for a MRC located in sandstone with ISRT less than 20 °C, the AT can be 7 passively controlled to not exceed 32 °C within 96 h due to the low-temperature surrounding rock (Zhang et al., 8 2018). For a MRC located in sandstone with an ISRT of 24 °C, the AT can be controlled below 32 °C within 96 h by 9 relying on the MCA of 0.3 m<sup>3</sup>/min per capita (Zhang et al., 2019). For a MRC with a slightly higher ISRT, MCA and 10 PCM cooling can be combined to achieve the temperature control goal. For a MRC with higher ISRT, technologies 11 such as ice storage cooling, MCA composited ice storage cooling, or surrounding rock pre-cooling composited MCA, 12 PCM cooling or ice storage cooling can be used to achieve the temperature control goal. However, the applicable 13 scope and economy of these methods need to be investigated.

# 14 **4.4** Acceptable range of environmental parameters

15 As far as the safety is concerned, it is permissible for personnel to be exposed to a CO concentration of 50 ppm for 8 h according to the Occupational Safety and Health Guideline for Carbon Monoxide. Although humans exposure 16 17 to a CO concentration of 400 ppm within 1 h may cause headaches, they dose not face the risk of death (Downs, 18 2016, Lee et al., 2020). It has been inferred that the CO concentration in a MRC caused by human metabolism is 19 below 50 ppm within 96 h, as a result, the environmental hazards caused by human metabolism of CO can be ignored. 20 Considering that harmful gases with high CO concentrations in the roadway may influx into the MRC with the entry 21 of people, in China, scrubbing measures must be equipped in the MRC to reduce the CO concentration from 400 ppm 22 to 24 ppm within 20 minutes. Fresh air ventilation is the most direct and effective measure to dilute harmful gases. 23 However, according to the calculation method (Zhang *et al.* 2020), when the per capita ventilation rate is  $0.3 \text{ m}^3/\text{min}$ , 24 it will take 30 minutes for MRC to reduce the CO concentration from 400 ppm to 24 ppm. If the requirement for rapid CO dilution within 20 minutes is met, the per capita ventilation rate should reach 0.45 m<sup>3</sup>/min. Therefore, 25 26 considering personnel safety and environmental control costs, it is recommended to increase the CO rapid dilution 27 time to 30 minutes, or increase the allowable value of CO concentration to 50 ppm within 20 minutes.

28 In personnel exposure environments, CO<sub>2</sub> gas is usually used as a symbolic gas for evaluating indoor air quality 29 (Zhu *et al.*, 2020), which affects human cognition and physiological safety. Although human is exposed to  $CO_2$  at 30 0.3% or 0.4% for several hours results in decreased cognitive performance (Kajtár *et al.*, 2012), exposed to CO<sub>2</sub> at 31 0.5% within 2.5 h does not effect on acute health symptoms of respiratory, visual and skin-related, headache, and 32 sensory (Zhang et al., 2016). However, exposure to  $CO_2$  at above 1% may lead to health symptoms of increased respiratory rate, respiratory acidosis, metabolic stress, increased brain blood flow, increased minute ventilation 33 34 (Azuma et al., 2018), and exposure to CO<sub>2</sub> at 1.2% with 32 °C WB and 85% RH led to significant headaches (Li et al., 2018). For CO<sub>2</sub> concentration in MRC, the allowable value in the current documents is 1%. Whereas under the 35 36 condition of a per capita ventilation rate of 0.3 m<sup>3</sup>/min, the CO<sub>2</sub> concentration in a MRC can be maintained to a low level of 0.3% (Zhang et al., 2020). Meanwhile, it is easy to reduce CO<sub>2</sub> concentration to below 0.5% through air 37 APDs, according to our experimental results. From the perspective of rescue, keeping conscious during the 38 39 evacuation period is more conducive to miners' judgment and escape. Therefore, it is recommended that the allowable 40 value of  $CO_2$  concentration can be limited to 0.5%, which will be acceptable in terms of economic costs.

The environment with AT above 32 °C and RH above 60% can be considered as a hot-humid environment (Shi *et al.*, 2013). The hot-humid environment has an important impact on human physiological health, in which the human body's heat dissipation is disrupted, leading to an increase in body temperature with a series of thermal symptoms and even death. Previous studies showed that exposure to RH at above 40% could be better for the eyes

and upper airways than levels below 30% (Wolkoff et al., 2007), and exposure to AT at 29 °C or below the heat 1 2 sensation is not obvious. However, when the AT is above 30 °C with RH 70%, some subjects felt uncomfortable 3 during 3 h exposure. When the AT reaches 32 °C, the human body feels warm, as the RH increases from 70% to 90%, 4 the heat sensation becomes more and more obvious (Jin et al., 2017). Hirata et al. (2015) found that young people 5 with an age of 20~30 exposed to the thermal environment with RH of 60% and temperature of 32.5 °C while standing 6 for 3 hours, the core temperature elevation in humans is 0.18 °C. In terms of the MRC where refugees will be 7 continuously exposed for above 96 h, to maintain allowable values of AT and RH are very important. However, 8 according to relevant documents of different countries, the allowable values in the MRC are different. In China, the 9 allowable values of AT and RH are 35 °C in dry bulb and 85%, respectively (Yuan et al., 2017). In the United States, the allowable value is apparent temperature 35 °C, namely, the AT is 28.5 °C when the RH is 85% (Halim et al., 10 11 2019). In Indonesia, the allowable values are 32 °C WB and 65% RH, respectively (Paul et al., 2019). Bernard et al. 12 (2018) observed that an apparent temperature of 49 °C could be sustained for 24 h based on the predicted heat strain 13 model. Ashley et al. (2019) found that sustainable exposures resulting in no significant increases in the physiological strain at an apparent temperature greater than 35 °C, with a likely ceiling below 46 °C AT. Hao et al. (2016) 14 15 recommended a safe upper limit of 35 °C apparent temperature for 96 h with an upper limit AT of 29.6 °C at 90% RH. Thus, it can be seen that in terms of thermal environment parameters in a MRC, the allowable value is relatively 16 17 moderate in Indonesia, but lower than that in the United States and higher than that in China. From the current results 18 and previous research, it is easy to control the RH in a MRC below 60%, but for MRCs with higher ISRT, maintaining 19 AT at a lower level means more cost. Considering personnel safety and cost, it is recommended that the RH and AT 20 of MRC be limited to 60% and 32 °C, respectively.

# 21 4.5 Recommended parameters for MRCs

Table 2 lists the recommended metabolic parameters and the environmental parameters for MRCs.

23 <b>T</b>	able 2. Metabolic parameter	s and environmental	parameters for MRCs
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Terms	Specific parameter	Recommend from others	Our recommend
Values of main	O <sub>2</sub> metabolic rate	0.29~0.37L/min for sitting men in Zhai et al.(2018)	0.33~0.37 L/min
metabolic	CO2 metabolic rate	0.23 to 0.34L/min in Yang et al. (2020)	0.34 L/min
parameter	Heat metabolic rate	113~134W in Bernard et al. (2018)	117~128 W
during refuge	Sweat metabolic rate	4.5g/min in Klein et al., (2017)	1 L/day
Acceptable	O <sub>2</sub> concentration	18.5%~23% in Halim et al., (2019)	18.5%~23%
values of	CO <sub>2</sub> concentration	1% in Halim et al., (2019)	0.5%
environmental	CO concentration	25ppm, from 400ppm to 25ppm within 20min in Halim	25 ppm, from 400 ppm to
parameters		<i>et al.</i> , (2019)	25 ppm within 30 min
	Ambient temperature	35°C apparent temperature in Halim et al., (2019), 32 °C	32°C WB and 65% RH
	and relative humidity	WB and 65% RH in Paul et al., (2019), 35°C WB and	
		85% RH in Yuan et al., (2017)	

# 24 5 Conclusions

22

In this research, an experiment of 50 subjects exposed in a MRC laboratory was carried out for 8.5 h to grasp the human metabolism during the refuge process and the characteristics of environmental control measures used in MRC. In addition, the possibility of passively scrubbing  $CO_2$  through  $Ca(OH)_2$  particle bags and the performance of dehumidification by MCA in the MRC were discussed by simulation experiments. Based on the results of these experimental studies, the following specific conclusions may be made.

(1) During the refuge period, the per capita metabolic rates of O<sub>2</sub>, CO<sub>2</sub> and heat, are 0.38 L/min, 0.34 L/min and
 130 W, respectively. The influence of trace gases such as CO and CH<sub>4</sub> produced by human metabolism on the indoor

1 MRC environment can be ignored.

2 (2) When  $Ca(OH)_2$  particles are used as  $CO_2$  adsorbents, the APD has the function of scrubbing  $CO_2$  and 3 dehumidification at the same time. Two devices can make the  $CO_2$  concentration below 1% and the RH below 85%, 4 and three devices can make the indoor  $CO_2$  concentration below 0.8% and RH below 76%. The influence of the 5 chemical heat caused by scrubbing  $CO_2$  and dehumidification on indoor AT should not be ignored. Although the 6 possibility of passively scrubbing  $CO_2$  exists, when the  $Ca(OH)_2$  particles are packaged to passively scrub  $CO_2$ , the 7 amount of the adsorbent will increase significantly for MRC.

(3) When MCA is used to dehumidification in a MRC, the capita air volume of 0.1 m<sup>3</sup>/min can control the RH
 below 80%, and 0.15 m<sup>3</sup>/min per capita can maintain the indoor RH close to 60%.

(4) In the early stage of disaster avoidance, the indoor AT will rise rapidly within 1 h, and then slowly rise, the
 surrounding rock parameters have a significant impact on the rise of AT. Some composite temperature control
 schemes have been proposed for MRC with different initial surrounding rock temperatures.

(5) Considering the safety of human exposure for 96 h and the economy of environmental control, it is
 recommended that, for MRC, the allowable values of the indoor environmental parameters of CO concentration, CO<sub>2</sub>
 concentration, AT and RH are 50 ppm, 0.5%, 32 °C, and 60%, respectively.

16 Due to the limitation of the experimental environment, although the test carried out in this article is similar to 17 the real situation of a MRC in terms of indoor air quality and humidity control. But for temperature control, there are 18 certain differences in thermal properties between the concrete wall of the MRC laboratory and the surrounding rock 19 of underground MRCs, also the variation of AT in the MRC laboratory may be affected by the external environment 20 in the later stage. Therefore, regarding to the study of MRC temperature control, the thermal response tests will be 21 conducted in an underground MRC in the future work. Apart from this, in terms of environmental protection in a 22 MRC, there are still many issues worthy of in-depth study, e.g., how to achieve a better passive control of  $CO_2$ 23 concentration through chemical adsorbents, how to improve the reliability and economy of cooling methods, how to 24 use the mine compressed air entering the MRC to rapidly control the indoor CO concentration. In the future work, 25 investigations on the above aspects will be performed to realize reliable and inexpensive environmental protection 26 technology for MRCs.

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